| 1 | Effect of Saccharomyces cerevisiae and Schizosaccharomyces pombe strains |
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| 2 | on chemical composition and sensory quality of ciders made from Finnish |
| 3 | apple cultivars |
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15 Abstract

16 Composition of volatile compounds and concentrations of sugars and organic acids were 17 studied in apple ciders produced with Saccharomyces cerevisiae and Schizosaccharomyces 18 pombe yeasts using eleven different Finnish apple cultivars. Moreover, sensory quality of 19 selected ciders was studied using check-all-that-apply test with untrained panelists. Seventy-20 seven volatile compounds were detected in the samples using HS-SPME-GC-MS. In general, 21 the ciders had higher concentrations of higher alcohols, aldehydes and acetals whereas the 22 juices contained higher C6-alcohols. Simultaneously, fermentation using S. pombe resulted in 23 lower concentrations of malic acid, ethyl pentanoate, ethyl hexanoate, and volatile acids and 24 higher concentrations of residual sugars compared to S. cerevisiae. Ciders made using S. 25 cerevisiae were characterized as 'alcoholic' and 'yeasty' while S. pombe ciders were more frequently described as 'sweet', 'honey-like', and less rated as sour. Besides the strong effect 26 27 by the yeasts, apple cultivars had significant effects on the compositional and sensorial 28 properties of apple ciders.

Keywords: Apple cultivar; Cider; *Schizosaccharomyces pombe*; Yeast; Volatile compounds;
Sensory quality.

32 **1. Introduction**

33 Apple (Malus domestica Borkh.) is the second highest consumed and the third most produced fruit in the world with an annual production of 86.2 million tons in 2018 reported by the Food 34 and Agriculture Organization Corporate Statistical Database (FAOSTAT, 2018). While the 35 36 majority (over 60%) of apple production are sold and consumed fresh, around 20% of them are 37 processed into value-added products, such as apple juice, jam, puree, cider, vinegar, and 38 dehydrated apple product (Shalini and Gupta, 2010). As a fermented alcoholic beverage, cider 39 possesses a complex aroma profile and its alcohol content ranges from 1.2% to 8.5% (ν/ν) 40 (Waston, 2013). Apple cider is one of the fastest growing beverages, the annual growth rate is 41 predicted to rise to 15% by 2021 in the alcoholic beverage markets, and its production gained 42 higher popularity in European countries, mainly in the United Kingdom, France, and Northern European countries (Jamir, Stelick & Dando, 2020; Miles, Alexander, Peck, Galinato, 43 44 Gottschalk & Nocker, 2020). According to the European Cider and Fruit Wine Association, around 7.6% of the overall apple crops in 2018 were used for cider production and 48.5% of 45 46 ciders were made from cider apples. Currently, there is increasing interest among cider producers in Northern European countries, especially in Finland, to develop the local cultivars 47 48 as cider apples. However, there are no breeding programs for speciality cider apples in Finland, 49 and no available data concerning on the potential of Finnish apples in cider production.

Flavor is one of the most important factors for assessing the olfactometric profile and quality of an alcoholic beverage (Symoneaux, Chollet, Patron, Bauduin, Le Quéré & Baron, 2015). The flavor of apple ciders is derived from complex mixtures of non-volatile compounds, like sugars, acids, phenolics, and volatile compounds, including esters, higher alcohols, fatty acids, aldehydes, ketones, terpenoids, and volatile phenols (Antón, Suárez Valles, García Hevia & Picinelli Lobo, 2014). Investigation of the volatile compounds in the final cider products is crucial, as the volatile compounds highly contribute to the sensory identities and flavor 57 properties. The content and profile of various sensory-active volatile compounds in ciders are 58 influenced by several factors, including apple varieties, growing conditions, ripeness, and yeast 59 strains involved in the fermentation process (Alberti et al., 2016; Braga et al., 2013; Rosend, 60 Kuldjärv, Rosenvald & Paalme, 2019). For example, ciders produced from dessert apple 61 cultivars may possess more desirable flavors and mouth-feel whereas cider apple cultivars 62 contain more phenolic compounds and are perceived more sour (Merwin, Valois & Padilla-63 Zakour, 2008). In addition to the cultivars, the ripeness of apples affects the composition of 64 volatiles. Ciders made from overripe apples contained more volatile compounds, such as higher 65 alcohols and acetate esters, than ciders made from less ripe apples (Braga et al., 2013; Rosend et al., 2019). Moreover, the addition of yeast assimilable nitrogen also enhanced the aromatic 66 67 complexity of cider, including ethyl and acetate ester concentrations, which has been 68 previously reported by Karl et al (2020).

69 Many of the volatile compounds in cider are derived from yeast metabolism during the 70 fermentation whereas some compounds originate from the apples (Pietrowski, Dos Santos, 71Sauer, Wosiacki & Nogueria, 2012). Numerous studies have reported that yeast species has an 72 important role in the volatile compounds of ciders (Albergaria and Arneborg, 2016; Benito, 73 2019; Luan, Zhang, Duan & Yan, 2018; Wei et al., 2020). Saccharomyces cerevisiae is 74 considered as the conventional commercial yeast species used for cider production due to its 75 strong fermentative power, high tolerance to harsh environment and low production of off-76 flavor compounds (Albergaria et al., 2016). Besides S. cerevisiae, the roles of other yeast in 77 Saccharomyces genus and various non-Saccharomyces yeasts (Schizosaccharomyces pombe, 78 Hanseniaspora opuntiae, Pichia kluyveri, and Pichia kudroavzevii) have been assessed in affecting the volatile compositions and enhancing the aroma complexity of ciders. S. pombe 79 80 has been suggested to be a potential alternative to Saccharomyces yeasts in fruit wine making 81 due to its equal fermentation capacity to the latter. This species also has been reported to reduce malic acid content and improve aroma complexity of alcoholic beverages compared to *S. cerevisiae* (Mylona et al., 2016). *S. pombe* has been used in apple wine studies (Satora, Semik-Szczurak, Tarko & Buldys, 2018), however, the available data regarding the influence of this
yeast on the volatile compounds of apple ciders is still scarce.

86 The main aim of the present study was to characterize the effects of S. cerevisiae and S. pombe 87 strains on the chemical compositions of apple cider products. These compositional analyses 88 included investigations of volatile compounds via headspace solid-phase microextraction 89 coupled with gas chromatography mass spectrometry (HS-SPME-GC-MS) and sugars and 90 acids using a GC coupled with flame ionization detector (GC-FID). Moreover, the sensory 91 properties of the selected cider samples were investigated with an untrained consumer panel. 92 Special focus was on the contribution of eleven Finnish apple cultivars (four summer cultivars, 93 five autumn cultivars, one winter cultivar, and one decorative cultivar) to the compositional 94 profiles of ciders. In Finland, early ripening apple crops are traditionally important due to its 95 short growth season, while late ripening apple cultivars contain more optimal conditions for 96 developing chemical and sensory qualities (Seppä, 2014). However, the Finnish apple cultivars 97 have not yet been characterized concerning their potential in cider production. Most of the 98 apple samples used in this study for cider fermentation belonged to old and traditional Finnish 99 cultivars without current significant commercial utilization as this current study aimed to 100 screen cultivars with fermentation potential.

101

102 **2. Materials and Methods**

103 2.1 Chemicals and strains

104 The culture of *Saccharomyces cerevisiae* Lalvin V1116 (V1116) was purchased from 105 Lallemand Inc. (Montreal, Canada) and the culture of *Schizosaccharomyce pombe* 3796

106 (SP3796) was provided by DSMZ Institute (Braunschweig, Germany). The volatile standards 107 of ethyl acetate, ethyl butanoate, methyl butanoate, 2-methylbutan-1-ol, 3-methylbutan-1-ol, 108 4-methylpentan-2-ol, 1-hexanol, 6-methylhept-5-en-2-one, hexanal, and hexane were 109 purchased from Sigma (St. Louis, MO), Aldrich (Gillingham, U.K.), and VWR (Darmstad, 110 Germany). For sugars and acids analysis, the standards of sucrose, glucose, fructose, sorbitol, 111 malic acid, quinic acid, succinic acid, and ascorbic acid were obtained from Extrasynthese 112 (Genay, France). An alkane mixture (C5-C30) was obtained from Sigma-Aldrich (St. Louis, MO). All the other solvents are of analytical grade. 113

114 2.2 Apples

115From the total of 97 Finnish seed born local apple cultivars (Malus x domestica Borkh.) 116 accepted to the Finnish national apple germplasm collection (Heinonen and Bitz, 2019), nine 117 apple cultivars, including 'Alasen Punainen' (AP), 'Kersti' (Kr), 'Lepaan Meloni' (LM), 'Lohjan Kirkas' (LK), 'Aino' (An), 'Gustavs Bästa' (GB), 'Luotsi' (Lt), 'Pieksämäki' (Pk), 118 119 and 'Turso' (Tr), were selected to the cider fermentation. In addition to the seed born local 120 apple cultivars, the Finnish cultivar 'Juuso' (Ju) from the Finnish breeding program was 121 included in the study. In the set selection of ten cultivated apples for this study, the criteria 122 were selected to obtain a variety of tastes, i.e. levels of sweetness, acidity, and bitterness (Supplementary Table 1). The one decorative apple cultivar (Malus 'Hyvingiensis', Hg) of 123 124 Finnish origin being in professional production was selected to get information about its 125suitability for cider production. Furthermore, the selected cultivars are in a nursery production 126 thus currently not in wider professional apple production in Finland.

127 The selected apples growing in the Finnish national apple germplasm collection in Piikkiö 128 (60°25'N, 22°31'E) in southwestern Finland were randomly harvested in the same orchard in 129 the Natural Resources Institute Finland (Luke). Subsequently, the over-ripeness program of 130 apples was carried out at 4 °C in darkness for one month. The completion of over-ripeness was determined by the iodine starch test (Supplementary Table 1). Eight ciders (four cultivars and
two yeasts) made from three apple cultivars 'Gustavs Bästa', 'Turso', and 'Juuso' and one
decorative apple cultivar 'Hyvingiensis' were included in the sensory analysis.

134 2.3 Cider preparation

Apples were washed with cold tap water and subsequently drained. Thereafter, the samples without obvious external damage were cut into small pieces and squeezed into juice with a centrifugal juice press (Vita Pro-Active JE810, Kenwood). The appearance or size of apples was not used as criteria in the sample selection. The apple juices were first pooled and divided into 50 mL glass tubes and then pasteurized in a hot water bath (95 °C) for 5 min and then cooled down to the room temperature in a cold-water bath.

141 The cider samples used for chemical analysis were produced by fermentations carried out in 142 100 mL Duran bottles with aliquots of 80 mL pasteurized apple juice. Prior to inoculation, the strains S. cerevisiae 1116 and S. pombe 3796 were proliferated in yeast extract-peptone-143 144 dextrose (YPD) liquid medium (1% yeast extract, 2% peptone, 2% dextrose) at 25 °C for 48 h 145with shaking at a speed of 150 rpm. The colony populations of S. cerevisiae 1116 and S. pombe 146 3796 were determined by spread plate technique after 48 h of propagation, respectively. Afterwards, yeast cells were collected from the broth and centrifuged at $4500 \times g$ for 10 min. 147 148 The pellets were washed using sterile solution of sodium chloride (0.9%) and centrifuged at $4500 \times g$ for 10 min, subsequently. The washing process was carried out twice successively. 149 After that, the yeast strains were resuspended in apple must for fermentation. All the cultures 150 were inoculated at the yeast cell amounts of 10⁷ CFU/mL. No optimization of fermentation 151152conditions was carried out in the current study. Fermentations were carried out in duplicates at a controlled temperature of 25 °C in darkness in an incubator (Memmert GmbH, Schwabach, 153Germany) and were considered to be completed when °Brix values reached constant levels and 154

155no CO₂ production from yeast growth during four consecutive days. The CO₂ production from 156 yeast growth was monitored every two days by calculating the weight loss of bottles. The °Brix 157 value was measured by a °Brix meter from Atago Co. Ltd. (Tokyo, Japan). The unfermented 158apple juices were set as the controls incubated under same conditions as the samples inoculated 159with yeast strains. After fermentation, the precipitates and/or yeast cells were removed from both juices and ciders by centrifugation at $3000 \times g$ for 10 min, and the obtained supernatants 160 161 were stored at -80 °C. However, the ciders used for sensory evolution were produced by 162 fermentations conducted in 500 mL Duran bottles using aliquots of 400 mL pasteurized apple 163 juice and stored at -20 °C. All the processes were conducted in food-grade conditions and 164 incubation of yeasts were carried out in aerobic conditions.

2.4 Measurements of pH, ethanol, and concentrations of sugars and organic acids during
 fermentation

167 The pH value was measured by a pH meter from WTW (Weilheim, Germany). Ethanol 168 concentration in the fermented apple ciders was determined by Shimadzu GC-2010plus gas 169 chromatograph (Shimadzu, Japan) coupled with flame ionization detector (FID) according to 170 a previous published method (Liu, Laaksonen, Marsol-Vall, Zhu & Yang, 2020). 171 Quantification of ethanol was carried out by using calibration curve of ethanol of different 172 concentrations (R^2 = 0.9997).

The determination of individual sugars and organic acids was carried out according to the GC-FID method described in a previous study with slight modifications (Liu, Laaksonen, Kortesniemi, Kalpio & Yang, 2018). In brief, individual sugars (sucrose, fructose, glucose, xylose, and sorbitol) as well as organic acids (malic acid, quinic acid, succinic acid, and ascorbic acid) were identified by comparing the retention times of individual compounds to those of their corresponding authentic standards. While these compounds were quantified using 179 calibration curves established by analysis of series of reference compounds of varying 180 concentrations. The total sugar and organic acid contents were calculated as the concentration 181 sum of concentrations of all individual sugars (sucrose, fructose, glucose, xylose, and sorbitol) 182 and organic acids (malic acid, quinic acid, succinic acid, and ascorbic acid), respectively.

183 2.5 Analysis of volatile compounds

184 The determination of volatiles in apple products was carried out using HS-SPME-GC-MS 185 according to a previous published method (Liu et al., 2020). Briefly, 2 mL of apple juice or cider, 0.2 g of sodium chloride, and 10 µL of internal standard (4-methylpentan-2-ol, 802 186 187 µg/mL in methanol) were mixed, placed in a 20 mL glass vial, and stored at 4 °C prior to use. 188 The extraction of volatile compounds from headspace was performed using a 2 cm SPME fiber 189 coated with a DVB/CAR/PDMS layer (50/30 µm, Supelco, Bellefonte, PA), and temperature 190 was kept at 45 °C for 30 min with agitation. The SPME fiber was set at 250 °C for 60 min 191 before the extraction process. The extracted compounds in the SPME fiber coating were 192 desorbed into a Trace 1310 (Thermo Scientific) gas chromatograph with a TSQ 8000 EVO 193 mass spectrometer (Thermo Scientific) in splitless mode at 240 °C for 3 min. The separation 194 of volatile compounds was carried out using a DB-WAX (J&W Scientific, Folsom, CA) 195 column (60 m \times 0.25 mm \times 0.25 μ m). A SPB-624 column (30 m \times 0.25 mm \times 0.25 μ m) from Agilent (Santa Clara, CA) was also used to assist the identification of volatiles. Helium was 196 197 chosen as the carrier gas and its flow rate was set at 1.6 mL/min. The oven temperature was set as 50 °C for 3 min followed by an increase to 200 °C at a rate of 5 °C/min and held at 200 °C 198 199 for 8 min. Mass spectra of analytes was recorded in electron impact (EI) ionization mode at 70 200 eV with a scan range from m/z 30 to m/z 300. The MS transfer line was set at 200 °C while the 201 ionization source temperature was at 220 °C. The identification of volatile compounds were 202 achieved by 1) comparing retention times and mass spectra between analytes and their 203 corresponding authentic standards, when available; 2) matching the obtained mass spectra of 204 analytes with those of the library (NIST 17) and comparing their retention indices (RIs) to the 205 data reported in the NIST Database (2017). Co-injection of a series of C5-C30 alkane mixture 206 under the same chromatographic conditions to that of apple samples was carried out to 207 determine the RI values of analytes. All the individual compounds identified by the DB-WAX 208 column were semi-quantified (relative concentrations) by comparing their base peak areas to 209 that of the internal standard, by using the following equation: $C (mg/L) = A_C * C_{LS} (mg/L) / A_{LS}$. 210 (C: relative concentration of analyte; C_{LS}: final concentration of internal standard in samples; A_C: peak area of analyte; A_{LS}.: peak area of internal standard). 211

212 2.6 Sensory evaluation

213 Voluntary untrained participants (n = 34, 26 females and 8 males, 20–65 years old) were asked 214 to evaluate the samples based on the check-all-that-apply (CATA) method, and the hedonic scale on appearance, odor (orthonasal), and flavor (including retronasal odor, taste, and 215 216 mouthfeel) as well as the intensities of color and taste by 9-point scales (Supplementary Table 217 3). CATA method was used to evaluate the characteristics of eight apple ciders (Varela and 218 Ares, 2012). A list of selected sensory attributes (n = 26, Supplementary Table 4) for apple 219 ciders were provided to the participants, and they were instructed to select all suitable 220 descriptors from the list or generate new attributes to describe the cider samples. The samples were served in balanced randomized order (Williams design) at room temperature à 10 mL in 221 222 50 mL transparent plastic cups covered with glass lids. Data was collected using Compusense 223 Cloud software (Compusense Inc., Guelph, Canada). The sensory evaluation was carried out 224 in a controlled sensory laboratory.

225 2.7 Statistical analysis

Variance analysis (ANOVA) and Tukey's (HSD) test were performed to evaluate the compositional differences among the different apple juices and ciders produced by two different yeast species using IBM SPSS Statistics 25.0 (SPSS Inc., Chicago, IL). Principal components analysis (PCA) and Partial least squares regression discrimination analysis (PLS-DA) were carried out to explore the differences among ciders produced from apples of different cultivars and among processing methods (juice, fermentation with *S. cerevisiae* 1116 and *S. pombe* 3796). Full cross validation was carried out in order to estimate a statistically reliable model. Multivariate method was carried out using Unscrambler X (Version 10.3, CAMO software, Oslo, Norway).

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236 **3. Results and discussion**

237 3.1 General fermentation parameters

The general fermentation characteristics, i.e. the total production of CO₂, ethanol contents, pH, 238 239 organic acids, and residual sugars, of apple juices and their corresponding cider samples are presented in Table 1. The fermentation kinetics varied depending on the different general 240 fermentation parameters (Belda, Navascués, Marquina, Santos, Calderon & Benito, 2015). The 241 242 fermentation carried out with the S. cerevisiae strain was completed after 16 days, while the 243 fermentation with S. pombe strain took 18–20 days. During fermentation, CO₂ was released in 244 a range from 3.33 to 6.51 g in the V1116 samples and 4.06 to 5.97 g of CO₂ in SP3796 samples. 245 The average alcohol content in V1116 ciders was 6.67% (ν/ν) whereas in SP3796 ciders was 246 6.45% (ν/ν). Ethanol yield was slightly influenced by the yeast strains, and the difference was 247 more dependent on the apple cultivars. For example, ciders made from 'Aino' had the lowest 248 ethanol contents among the cider samples (5.50% for V1116 cider and 5.19% for SP3796 cider). 249 The low ethanol yield of fresh apple juice without additional sugars and acids has been also 250 reported in the previous studies (Pando Bedriñana et al., 2017; Rumpunen, Ekholm, & Nybom, 251 2017).

252Similar to the CO₂ and ethanol productions, pH values were significantly influenced by apple 253cultivars in both apple juices and ciders (Table 1). Organic acid composition in apple juice is also highly cultivar dependent. The highest amount of malic acid, the major organic acid in 254 255 apples (Lee and Wrolstad, 1988), was detected in cultivar 'Aino' (18.94 g/L), followed by cultivars 'Juuso' (11.31 g/L) and 'Turso' (9.58 g/L). The 'Aino' ciders had the lowest pH 256 257 among the cultivars, and the 'Pieksämäki' ciders contained the highest pH values. The pH 258 values were significantly higher in the ciders produced by S. pombe compared to the 259 corresponding juices prepared from apples of the same cultivars. However, there was no 260 significant change in pH values before and after fermentation with S. cerevisiae. The differences could be explained by high malic acid consumption of S. pombe (Benito, 2019; Liu 261 262 et al., 2018), as 47-89% of malic acid was consumed in the fermentations with S. pombe. In 263 contrast, the corresponding decrease was only 6-18% resulting from fermentation with S. 264 cerevisiae as shown in Table 1. The difference between the two yeast strains is also shown 265 along the second PC in the PCA model (Supplementary Figure 1). The PC2 also separated the 266 cultivar 'Aino' (An) from the others. After fermentation, the content of quinic acid decreased slightly, whereas ascorbic acid and succinic acid, which were absent in apple juices, became 267 268 detectable in ciders. The reduction of organic acids with the ciders produced from S. pombe, especially malic acid, and therefore the increase of pH values may mitigate the flavor of harsh 269 270 green apple sourness, acidity, and puckering astringency (Laaksonen, Mäkilä, Tahvonen, 271 Kallio & Yang, 2013).

With regard to sugars, fructose was the main sugar presented in apple juices, followed by sucrose, glucose, and sorbitol (Table 1). These results are in accordance with those reported by Pires et al (2018). Apple cultivars significantly influenced the contents of sugars, the highest amounts of fructose and glucose were detected in cultivar 'Pieksämäki' (108.10 g/L) and 'Hyvingiensis' (52.85 g/L), respectively. The concentration of sugars decreased dramatically after fermentation with more than 90% of sugars consumed. Fermentations with *S. pombe* 3796 consumed less sugar in comparison to those with the *S. cerevisiae* V1116. The difference in consumption of sugars between those two yeasts could be ascribed to the metabolic characteristics of the yeasts (Liu et al., 2018). The differences in sugars were also presented in the PCA model (Supplementary Figure 1), which showed a clear difference between the initial juice samples and fermented samples.

283 *3.2 Volatile compounds in apple juices and ciders*

284 A total of 77 volatiles were identified in apple juices and their corresponding fermented samples by the DB-WAX Column including 34 esters, 5 aldehydes, 20 higher alcohols, 4 285 286 ketones, 1 terpene, 3 acetals, 1 hydrocarbon, 1 oxide, 2 benzenes, and 6 acids (Table 2). Their concentrations in apple juices and completely fermented apple ciders are listed in 287 288 Supplementary Table 2, whereas the total contents of each volatile compound class are 289 summarized in Table 3. Twenty-nine compounds, mainly ethyl esters and higher alcohols, were 290 detected only in cider samples, while 20 compounds, mainly acetate esters and other esters 291 present in the juices, became undetectable after fermentation.

292 Esters dominated in numbers among all classes of volatile compounds in the ciders, including ethyl esters, acetate esters, and other esters (Table 3). Although some esters originally existed 293 294 in the apple juices, most of them were produced during fermentation due to the esterification 295 of alcohols (Peng, Li, Cui & Guo, 2015). Ethyl acetate was the major ester in ciders 296 (Supplementary Table 2). It was reported to contribute to *pineapple*, *sweet*, or *pungent* order 297 (Li, Zhao, Zuo, Zhang, Zhang & Chen, 2020). Fermentation with S. cerevisiae yeasts produced 298 more ethyl acetate than that with S. pombe yeasts. This is in line with report in wine production (Peinado, Moreno, Maestre, Ortega, Medina & Mauricio, 2004). Among the other minor esters, 299 300 ethyl butanoate, ethyl decanoate, 3-methylbutyl acetate, phenethyl acetate, and hexyl acetate

were the quantitatively predominant fractions of volatile compounds of apple ciders, the results
 are in agreement with a previous study (Ye, Yue & Yuan, 2014).

303 Alcohols were the most abundant principal volatile group in the apple ciders in terms of 304 contents (Table 3). Higher alcohols are regarded as one of the most important precursors of esters, but they may have adverse effects on the quality of final products at excessive levels 305 306 (Satora, Cioch, Tarko & Wołkowicz, 2016). Most of the higher alcohols were produced during 307 yeast fermentation, the contents of higher alcohols in ciders were 7-25 times higher than those 308 in the initial juices. Apple ciders fermented by S. cerevisiae strain were more abundant with higher alcohols than those fermented by S. pombe yeasts. 2-Methylpropan-1-ol, butan-1-ol, 3-309 methylbutan-1-ol, pentan-1-ol, 1-hexanol, 6-methyl-5-hepten-2-ol, 2-ethylhexan-1-ol, and 310 pentadecane-8-ol were found in all the apple cider samples. 3-Methylbutan-1-ol is the most 311 312 predominant higher alcohols according to Supplementary Table 2, similar results have also 313 been reported in Chinese, French, and Danish apple ciders (Fan, Xu & Han, 2011, Qin, Petersen 314 & Bredie, 2018, Villière, Arvisenet, Lethuaut, Prost & Sérot, 2012). 2-Methylbutan-1-ol was 315 identified in initial apple juices, but it showed a significant decrease during yeast fermentation. 316 Butan-1-ol and pentadecane-8-ol showed relatively high concentrations in all cider samples 317 compared to the corresponding apple juices (Supplementary Table 2), and their concentrations 318 were dependent on both cultivar and yeast. Although C6-alcohols have been reported to be 319 very important alcohols in apple juices, they may have a negative effect on apple cider at 320 excessive levels (Qin et al., 2018). Among C6-alcohols identified in the current work, 1hexanol existed in original apple juices, and its concentrations decreased during fermentations. 321 322 (E)-2-Hexen-1-ol was only present in 'Gustavs Bästa' (GB), 'Luotsi' (Lt), and 'Pieksämäki' (Pk) juices, and was not detected in cider samples (Supplementary Table 2). 323

Five aldehydes and four ketones were identified in the cider samples and their corresponding juices (Table 2). Acetaldehyde was the most abundant aldehyde in initial juices and fermented 326 samples (Supplementary Table 2). This compound has been reported as a contributor of *fruity* 327 and *nutty* ordors, however, a high concentration of acetaldehyde (> 110 mg/L) may contribute 328 to *pungent* and *ether* aroma (Li et al., 2020, Luan, Zhang, Duan & Yan, 2018). Notably high 329 levels of acetaldehyde were detected in the ciders made from cultivar 'Lepaan Meloni' (LM), 'Lohjan Kirkas' (LK), and 'Gustavs Bästa' (GB), indicating that the accumulation of 330 331 acetaldehyde in ciders is apple cultivar dependent. Butanal, hexanal, and (E)-hex-2-enal were 332 detected in apple juices, and they disappeared after fermentations. Higher levels of 3-333 hydroxybutan-2-one (acetoin) were detected in the ciders produced from cultivar 'Juuso' (Ju). 334 The other three ketones, 4-methylpentan-2-one, 6-methylhept-5-en-2-one, and 2,6,8-335 trimethylnonan-4-one, were found in low concentrations from both apple juices and ciders.

336 Six volatile acids (acetic acid, 2-methylpropanoic acid, butanoic acid, 3-methylbutanoic acid, 337 pentanoic acid, and hexanoic acid) were identified in apple ciders. Volatile acids are important 338 for aromatic complexity in apple ciders and they can contribute to vinegar-like, sweat, and 339 rancid notes of apple ciders (Qin et al., 2018). Acetic acid was found at high contents in all the 340 cider samples in this work. Acetic acid is formed from oxidation of acetaldehyde during 341 alcoholic fermentation potentially contributing to sour and vinegar-like odors (Niu, Wang, Xiao, Zhu, Sun & Wang, 2019). The highest content of acetic acid was found in the samples 342 fermented with V1116 (Supplementary Table 2). This result is in accordance with a report 343 concerning apple ciders fermented with Saccharomyces and non-Saccharomyces yeasts 344 (Madrera, Lobo & Alonso, 2010). 345

The difference between the apple juices and their corresponding cider samples (Y-data, n=3) in the volatile composition (X-data, n = 77) was analyzed using PLS-DA (Figure 1A). In the PLS model with five validated factors ($R^2 = 0.959$; validated $R^2 = 0.913$), the apple juices can be separated clearly from the cider samples already on the first factor. Apple juices are located on the right side of factor-1 with strong positive correlation with 2-methylbutan-1-ol (HA_3), 351 hexyl butanoate (OE_6), hexyl 2-methylbutanoate (OE_7), hexanal (Ad_4), 4-methylpentan-352 2-one (K_1), and 1-hexanol (HA_9). After fermentation, the total concentrations of each group of volatile compounds elevated significantly. The separation of the cider samples fermented 353 354 with V1116 from those with SP3796 were shown in another PLS-DA model with three validated factors (Figure 1B; $R^2 = 0.972$; validated $R^2 = 0.937$). High production of several 355 356 ethyl esters, such as ethyl pentanoate (E_6) and ethyl hexanoate (E_7), and volatile acids, such as 2-methylpropanoic (Ai_2), 3-methylbutanoic (Ai_4), pentanoic (Ai_5), and hexanoic (Ai_6) 357 358 acids, separated V1116 samples clearly from SP3796 samples on the first two factors. At the 359 same time, clear differences were detected among the cultivars along the factor-1, as the cultivars 'Lohjan Kirkas' (LK) and 'Gustavs Bästa' (GB) characterized by higher concentration 360 361 of acetals, such as 1-ethpxy-1-methoxyethane (AC_1), 1,1-diethoxythane (AC_2), and 1-(1-362 ethoxyethoxy) pentane (AC_3), acetate esters, such as 3-methylbutyl acetate (AE_5) and 363 phenethyl acetate (AE_9), higher alcohols, such as 3-methylpentan-1-ol (HA_8), 2-ethylhexan-364 1-ol (HA_16), and pentadecan-8-ol (HA_20), acetic acid (Ai_1), and acetaldehyde (Ad_1). 365 Although the apple cultivars influenced the volatile compositions, the major difference came from the yeast strains according to the PLS-DA classification. A total of 51 volatile compounds 366 were detected in the ciders fermented with S. cerevisiae while 54 volatile compounds were 367 detected in the cider samples fermented with S. pombe. Thus, compared to the commercial S. 368 369 cerevisiae yeast, non-Saccharomyces yeasts can produce different sets of volatile compounds. 370 However, fermentation with SP3796 produced less ethyl esters, higher alcohols, and volatile acids compared to that with V1116 according to Table 3. The results are in accordance with 371 the findings in bilberry and grape wines (Liu et al., 2020; Peinado et al., 2004). 372

373 *3.3 Effect of apple cultivars on the volatile composition of apple ciders*

Principal component analysis (PCA) models were applied to visualize the relationships between volatile compound profiles and apple cultivars within the three sample types. As 376 shown in Figure 2A with only juice samples, the first PC separated apple juices according to 377 the apple cultivars. Juices made from cultivar 'Juuso' (number 10) located on the left side whereas the others on the right of PC1. This cultivar correlated with certain acetate esters 378 379 (AE 2, propyl acetate; AE 4, butyl acetate; AE 7, hexyl acetate; AE 8, octyl acetate), other esters (OE_1, methyl butanoate; OE_11, diethyl benzene-1, 2-dicarboxylate), and higher 380 381 alcohols (HA_12, 1-octen-3-ol; HA_14, heptan-1-ol; HA_18, octan-1-ol). The second PC also 382 discriminated some of the juice samples on the basis of their cultivars as the summer cultivars 383 ('Alasen Punainen', 'Kersti', 'Lepaan Meloni', and 'Lohjan Kirkas') were located on the 384 negative side of PC2, and the others on the positive side. On the PC2, variables ethyl 2methylpropanoate (E_3) and 2-methylbutanoate (E_5), hexyl 2-methylpropanoate (OE_3) and 385 386 2-methylbutan-1-ol (HA_3) correlated with the summer apple cultivars, while variables butyl 387 and hexyl butanoate (OE_2 and OE_6, respectively), (E)-2-hexen-1-ol (HA_11), and 4-388 methylpentan-2-one (K_1) were linked to the juices of the autumn apple cultivars.

389 In Figures 2B and 2C with the cider samples fermented with S. cerevisiae and S. pombe, the 390 first PCs separated the ciders made from cultivars 'Lepaan Meloni' (LM), 'Lohjan Kirkas' 391 (LK), and 'Gustavs Bästa' (GB) from those made from apples of other cultivars due to the 392 difference in concentration of acetate esters and higher alcohols. The second PC in the model 393 for S. cerevisiae samples (Figure 2B), separated the cultivars 'Lohjan Kirkas' (LK) and 394 'Gustavs Bästa' (GB), but they were not separated in the model based on the ciders fermented 395 with the S. pombe samples (Figure 2C). The PC2 in Figure 2C separated also cultivars 'Aino' 396 (An) and 'Luotsi' (Lt) from the others; ciders of both cultivars correlated positively with 2-397 methylpropan-1-ol (HA_1) and negatively with methyl acetate (AE_1). Similar separation was not detected with the cider samples fermented with V1116 (Figure 2B). The volatile compound 398 399 profiles were more influenced by the fermentation process in comparison to the cultivar 400 differences. Moreover, the PCA results for apple ciders (Figure 2A-C) shared similar patterns

with juices among apple cultivars were observed. This might be resulted from the different
compositional difference in primary volatile compounds among the cultivars indicating the
cultivar difference (cultivar X) as a potential reason for variation in this study, which has been
investigated also in previous apple cider studies (Alberti et al, 2016; Braga et al., 2013; Rosend
et al., 2019).

406 *3.4 Sensory quality of cider samples*

407 Sensory quality of selected eight cider samples, four fermented from each of the two yeast strains, were characterized using the rated sensory attributes and CATA descriptors to highlight 408 409 the potential differences between the yeasts. As shown in the PCA model in Figure 3A, a clear 410 differentiation based on the yeast strain was observed on the PC1 among the ciders produced 411 from cultivar 'Turso' (Tr), 'Juuso' (Ju), and 'Hyvingiensis' (Hg). The ciders produced by 412 V1116 were rated higher in appearance liking and the intensities of sourness, puckering 413 astringency, and mouth-drying astringency. These results were in line with the higher 414 consumption of malic acids by SP3796 yeast strain than V1116. However, the average ratings 415 were similar between the yeasts (Supplementary Table 3) as only clear differences were 416 observed in sourness and puckering astringency. Interestingly, the perceived bitterness of the 417 ciders was similar between the yeast strains (Supplementary Table 3). This attribute is typically linked to the non-volatile phenolic compounds, which were not investigated in this study. 418 419 Potentially, the reduction of acidity using S. pombe strains may result in higher intensities of 420 other attributes, e.g. bitterness. Moreover, the ciders produced from 'Gustavs Bästa' (GB) 421 behaved differently in comparison to the other cultivars. They were rated as less sour and 422 astringent compared to others. This effect can be explained by the lower organic acid contents 423 and higher sugar contents in this cultivar (Table 1).

424 The most often selected CATA descriptors were 'fruity', 'cider-like', 'fermented', 'sweet', and 425 'floral' (Supplementary Table 4), which were similar with a previous study of commercial 426 apple ciders (Qin et al., 2018). Among the list of 26 descriptors, only eight attributes resulted 427 in significant differences between the samples based on Cochran's Q-test (p < 0.05). In a PCA 428 model using the frequencies of the CATA descriptors (Figure 3B), the samples made with 429 different yeasts were clearly separated along the PC1 based on the CATA attribute variables. In the loadings plot, the samples fermented with V1116 on the right, can be described as 'sharp', 430 431 'dry apple', 'alcoholic', 'cider-like', 'yeasty', 'earthy', 'fermented', 'acidic', and 'cooked 432 apple'. Ciders fermented with SP3796 located on the left side and were mainly characterized as 'floral', 'fruity', 'tropical fruity', 'honey', 'sweet', and 'diverse'. The improvement on 433 434'fruity' odor and mouth-feeling from inoculation with non-Saccharomyces yeasts has also been 435 detected in apple cider and fruit wine productions (Magalhães et al., 2017; Varela, 2016). 436 Additionally, the second PC separated the cultivar 'Gustavs Bästa' (GB) from the others, being characterized as more 'spicy', 'cooked apple', and 'diverse'. 437

438 **4. Conclusion**

439 In conclusion, the effect of two different yeast strains (S. cerevisiae and S. pombe) on the 440 chemical composition of ciders was studied using 11 Finnish domestic apple cultivars. Both fermentation processes resulted in sharp increases in the contents of volatile compounds, 441 442 especially esters, higher alcohols, aldehydes, acetals, and acids. The major differences between 443 cider samples originated from fermentation with these two yeast species, although apple 444 cultivars had significant impact on the final chemical composition. Certain apple cultivars, 445 including 'Lepaan Meloni' (LM), 'Lohjan Kirkas' (LK), and 'Gustavs Bästa' (GB), were clearly separated from the others using multivariate models based on their volatile 446 compositions or non-volatile acid and sugar profiles. Fermentations with different yeast strains 447

also affected sensory properties of the ciders. Use of S. pombe in ciders generally decreased 448 449 the amounts of malic acid and lead to a decrease of sourness in the cider product. However, this effect may also result in the higher intensities of other attributes, such as bitterness. 450 451 Furthermore, more studies are needed to find out the impact of fermentation with S. pombe 452 strains on the phenolic composition of ciders and thus their contribution to the 'bitterness' and 453 'astringent' properties. At the same time, these ciders were described as more 'floral', 'fruit', and 'sweet' in comparison to the S. cerevisiae ciders, which were typically more 'cider-like', 454455 and 'alcoholic'. This study demonstrates the potential of using S. pombe strains in cider 456 processing. Furthermore, the study promotes the exploitation of old, local, traditional apple 457 cultivars, which are not currently utilized commercially, in cider processing.

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463 **Figure Captions**

Figure 1. PLS-DA models of volatiles as X-data (n = 77) to illustrate the differences between apple juices and apple ciders (A) or between the ciders (B) produced by fermentation with two different yeast strains. *S. cerevisiae* 1116 red circles (V, n = 44), *S. pombe* 3796 blue rectangles (SP, n = 44), juices green triangle (J, n = 22). Abbreviations refer to apple cultivars in Supplementary Table 1 and variables refer to Table 2.

469 **Figure 2.** PCA plots of volatile compositions in apple juices (A: Juice, 44 volatiles, 22 samples)

470 and apple ciders (B: S. cerevisiae 1116, 51 volatiles, 44 samples; C: S. pombe 3796, 54 volatiles,

- 471 44 samples) produced with different apple cultivars. Abbreviations refer to apple cultivars in
- 472 Supplementary Table 1 and variables refer to Table 2.
- 473 **Figure 3.** PCA models for sensory attributes of eight apple cider samples. **A**. rated pleasantness
- 474 and sensory attributes, n = 11 (Supplementary Table 3), **B**. check-all-that-apply attributes, n =
- 475 **26**.

476 Supporting Information Description:

477 **Supplementary Table 1.** Description of the apple cultivars used in this study

- 478 Supplementary Table 2. Semi-quantification of volatiles (mg/L) in the apple ciders fermented
- 479 with S. cerevisiae 1116 and S. pombe 3796 and their corresponding juices
- 480 **Supplementary Table 3.** Mean and standard deviations of liking and sensory attributes rated

481 on scales 1-9.

- 482 Supplementary Table 4. Sensory descriptors used in the CATA test and their frequencies (n)
 483 presented in the order of most often selected attributes.
- Supplementary Figure 1. PCA models for sugar and acid compounds in juice and cider
 samples (n = 110; two analytical replicates from 55 samples). Juices blue rectangles (n = 22), *S. cerevisiae* 1116 green triangles (n = 44), *S. pombe* 3796 red circles (n = 44). Abbreviations
- in the loading plot refer to apple cultivars in Supplementary Table 1.

488

490 **References**

- 491 Albergaria, H., & Arneborg, N. (2016). Dominance of Saccharomyces Cerevisiae in Alcoholic
- 492 Fermentation Processes: Role of Physiological Fitness and Microbial Interactions.
- 493 Applied Microbiology and Biotechnology, 100(5): 2035-2046.
 494 <u>http://doi.org/10.1007/s00253-015-7255-0</u>.
- 495 Alberti, A., dos Santos, T. P. M., Zielinski, A. A. F., dos Santos, C. M. E., Braga, C. M., ...,
- 496 Nogueira, A. (2016). Impact on Chemical Profile in Apple Juice and Cider Made from
- 497 Unripe, Ripe and Senescent Dessert Varieties. *LWT-Food Science and Technology*, 65:
- 498 436-443. https://doi.org/10.1016/j.lwt.2015.08.045.
- 499 Antón, M. J., Suárez Valles, B., García Hevia, A., & Picinelli Lobo, A. (2014). Aromatic 500 Profile of Ciders by Chemical Quantitative, Gas Chromatography-Olfactometry, and 501 Sensory Analysis. Journal Science. 79 (1), S92–S99. of Food 502 https://doi.org/10.1111/1750-3841.12323.
- Belda, Ignacio., Navascués, E., Marquina, D., Santos, A., Calderon, F., & Benito, S. (2015).
 Dynamic Analysis of Physiological Properties of *Torulaspora Delbrueckii* in Wine
 Fermentations and its Incidence on Wine Quality. *Applied Microbial and Cell Physiology*,
 99 (4): 1911-1922. http://doi.org/10.1007/s00253-014-6197-2.
- Benito, S. (2019). The Impacts of Schizosaccharomyces on Winemaking. Applied
 Microbiology and Biotechnology, 103(11): 4291-4312. http://doi.org/10.1007/s00253019-09827-7.
- 510 Braga, C. M., Zielinski, A. A. F., da Silva, K. M., de Souza, F. K. F., Pietrowski, G. A. M.,,
- 511 Nogueira, A. (2013). Classification of Juices and Fermented Beverages Made from Unripe,
- 512 Ripe and Senescent Apples Based on the Aromatic Profile Using Chemometrics. *Food*
- 513 *chemistry*, 141(2): 967-974. https://doi.org/10.1016/j.foodchem.2013.04.007.

- 514 Fan, W., Xu, Y., & Han, Y. (2011). Quantification of Volatile Compounds in Chinese Ciders
- 515 by Stir Bar Sorptive Extraction (SBSE) and Gas Chromatography-Mass Spectrometry
- 516 (GC-MS). Journal of the Institute of Brewing, 117(1): 61-66.
 517 https://doi.org/10.1002/j.2050-0416.2011.tb00444.x.
- FAOSTAT. (2018). Food and Agriculture Organization Corporate Statistical Database. FAO
 Online Database. Available online: http://www.fao.org/faostat/en/# data (accessed on 20
 October 2020).
- Heinonen, M., & Bitz, L. (2019). How to Discover Traditional Varieties and Shape in a
 National Germplasm Collection: The Case of Finnish Seed Born Apples (Malus x
 domestica Borkh.). *Sustainability*, 11(24), 7000. https://doi.org/10.3390/su11247000.
- Jamir, S. M. R., Stelick, A., & Dando, R. (2020). Cross-cultural examination of a product of
 differing familiarity (Hard Cider) by American and Chinese panelists using rapid profiling
 techniques. *Food Quality and Preference*, 79, 103783.
 https://doi.org/10.1016/j.foodqual.2019.103783.
- 528 Karl, A. D., Brown, M. G., Ma, S., Sandbrook, A., Stewart, A. C., Cheng, L., ... & Peck, G. M. 529 (2020). Foliar urea applications increase yeast assimilable nitrogen concentration and alcoholic 530 fermentation in 'Red Spy' used for cider rate apples production. HortScience, 55(8), 1356-1364. https://doi.org/10.21273/HORTSCI15029-531532 20.
- Laaksonen, O., Mäkilä, L., Tahvonen, R., Kallio, H., & Yang, B. (2013). Sensory Quality and
 Compositional Characteristics of Blackcurrant Juices Produced by Different Processes.
 Food chemistry, 138 (4): 2421-2429. https://doi.org/10.1016/j.foodchem.2012.12.035.
- 536 Lee, H. S., & Wrolstad, R. E. (1988). Apple Juice Composition: Sugar, Nonvolatile Acid, and
- 537 Phenolic Profiles. Journal of the Association of Official Analytical Chemists, 71(4): 789-
- 538 794. https://doi.org/10.1093/jaoac/71.4.789.

- Li, C. X., Zhao, X. H., Zuo, W. F., Zhang, T. L., Zhang, Z. Y., & Chen, X. S. (2020). The
 Effects of Simultaneous and Sequential Inoculation of Yeast and Autochthonous *Oenococcus Oeni* on the Chemical Composition of Red-fleshed Apple Cider. *LWT-Food Science and Technology*, 124: 109-184. https://doi.org/10.1016/j.lwt.2020.109184.
- Liu, S., Laaksonen, O., Marsol-Vall, A., Zhu, B., & Yang, B. (2020). Comparison of volatile
 composition between alcoholic bilberry beverages fermented with non-saccharomyces
 yeasts and dynamic changes in volatile compounds during fermentation. *Journal of Agricultural and Food Chemistry*, 68(11): 3626-3637.
 https://doi.org/10.1021/acs.jafc.0c01050.
- Liu, S., Laaksonen, O., Kortesniemi, M., Kalpio, M., & Yang, B. (2018). Chemical 548 549 Composition of Bilberry Wine Fermented with Non-Saccharomyces Yeasts (Torulaspora 550 Delbrueckii and Schizosaccharomyces Pombe) and Saccharomyces Cerevisiae in Pure, Fermentations. 551 Sequential and Mixed Food chemistry, 266: 262-274. 552 https://doi.org/10.1016/j.foodchem.2018.06.003.
- Luan, Yu., Zhang, B. Q., Duan, C. Q., & Yan, G. L. (2018). Effects of Different Pre-
- 554 fermentation Cold Maceration Time on Aroma Compounds of Saccharomyces Cerevisiae
- 555 Co-fermentation with *Hanseniaspora Opuntiae* or *Pichia Kudriavzevii*. *LWT-Food* 556 *Science and Technology*, 92: 177-186. https://doi.org/10.1016/j.lwt.2018.02.004.
- Madrera, R. R., Lobo, A. P., & Alonso, J. J. M. (2010). Effect of Cider Maturation on the
 Chemical and Sensory Characteristics of Fresh Cider Spirits. *Food Research International*,
- 559 43(1): 70-78. https://doi.org/10.1016/j.foodres.2009.08.014.
- Merwin, I. A., Valois, S., & Padilla-Zakour, O. I. (2008). Cider apples and cider-making
 techniques in Europe and North America. *Horticultural Reviews*, 34: 365.
- 562 Miles, C. A., Alexander, T. R., Peck, G., Galinato, S. P., Gottschalk, C., & Nocker, S. Van.
- 563 (2020). Growing Apples for Hard Cider Production in the United States—Trends and

 564
 Research
 Opportunities.
 HortTechnology,
 30
 (2),
 148–155.

 565
 http://doi.org/10.21273/HORTTECH04488-19.
 565
 148–155.

566 Mylona, A. E., Del Fresno, J. M., Palomero, F., Loira, I., Bañuelos, M. A.,, Suárez-Lepe,

- 567 J. A. (2016). Use of *Schizosaccharomyces* Strains for Wine Fermentation-Effect on the
- 568 Wine Composition and Food Safety. *International Journal of Food Microbiology*, 232:
- 569 63-72. https://doi.org/10.1016/j.ijfoodmicro.2016.05.023.
- National Institute of Standards and Technology. (2017). NIST Chemistry WebBook, Standard
 Reference Database Number 69. https://webbook.nist.gov/chemistry/name-ser/. Accessed
 25th August 2017.
- Niu, Y., Wang, P., Xiao, Z., Zhu, J., Sun, X., & Wang, R. (2019). Evaluation of the Perceptual
 Interaction Among Ester Aroma Compounds in Cherry Wines by GC–MS, GC–O, Odor
 Threshold and Sensory Analysis: An Insight at the Molecular Level. *Food chemistry*, 275:

576 143-153. https://doi.org/10.1016/j.foodchem.2018.09.102.

- Pando Bedriñana, R., Mangas Alonso, J. J., & Suárez Valles, B. (2017). Evaluation of
 autochthonous Saccharomyces bayanus strains under stress conditions for making ice
 ciders. LWT Food Science and Technology, 81, 217–225.
 https://doi.org/10.1016/j.lwt.2017.03.055.
- Peng, B., Li, F., Cui, L., & Guo, Y. (2015). Effects of Fermentation Temperature on Key Aroma
 Compounds and Sensory Properties of Apple Wine. *Journal of Food Science*, 80(12):
 S2937-S2943. https://doi.org/10.1111/1750-3841.13111.
- Peinado, R. A., Moreno, J. J., Maestre, O., Ortega, J. M., Medina, M., & Mauricio, J. C. (2004).
- 585 Gluconic Acid Consumption in Wines by *Schizosaccharomyces Pombe* and its Effect on
- 586 the Concentrations of Major Volatile Compounds and Polyols. *Journal of Agricultural*
- 587 and Food Chemistry, 52 (3): 493-497. https://doi.org/10.1021/jf035030a.

- 588 Pietrowski, G. A. M., Dos Santos, C. M. E., Sauer, E., Wosiacki, G., & Nogueria, A. (2012).
- Influence of fermentation with *Hanseniaspora sp.* Yeast on the volatile profile of fermented apple. *Journal of Agricultural and Food Chemistry*, 60 (39), 9815-9821.
- 591 https://doi.org/10.1021/jf302290k.
- ⁵⁹² Pires, T. C. S. P., Dias, M. I., Barros, L., Alvesad, M. J., Oliveirac, M. B. P. P., ..., Ferreira,
- 593 C.F.R. (2018). Antioxidant and Antimicrobial Properties of Dried Portuguese Apple
- Variety (Malus Domestica Borkh. cv Bravo De Esmolfe). *Food chemistry*, 240: 701-706.
 https://doi.org/10.1016/j.foodchem.2017.08.010.
- ⁵⁹⁶ Qin, Z., Petersen, M. A., & Bredie, W. L. P. (2018). Flavor Profiling of Apple Ciders from the
- 597 UK and Scandinavian Region. *Food Research International*, 105: 713-723.
 598 https://doi.org/10.1016/j.foodres.2017.12.003.
- Rosend, J., Kuldjärv, R., Rosenvald, S., & Paalme, T. (2019). The Effects of Apple Variety,
 Ripening Stage, and Yeast Strain on the Volatile Composition of Apple Cider. *Heliyon*,
 5(6): 01953. https://doi.org/10.1016/j.heliyon.2019.e01953.
- Rumpunen, K., Ekholm, A., & Nybom, H. (2017). Swedish apple cultivars vary in traits for
- juice and cider making. *Acta Horticulturae*, *1172*, 255–258.
- 604 https://doi.org/10.17660/ActaHortic.2017.1172.48.
- Satora, P., Cioch, M., Tarko, T., & Wołkowicz, J. (2016). Killer Strains of *Saccharomyces*:
 Application for Apple Wine Production. *Journal of the Institute of Brewing*, 122(3): 412421. https://doi.org/10.1002/jib.338.
- Satora, P., Semik-Szczurak, D., Tarko, T., & Buldys, A. (2018). Influence of selected
 Saccharomyces and *Schizosaccharomyces* strains and their mixed cultures on chemical
- 610 composition of apple wines. Journal of Food Science, 83 (2), 424-
- 611 431. https://doi.org/10.1111/1750-3841.14042.

- Seppä, L. (2014). Domestic Apple Cultivars Sensory Descriptions and Consumer Responses.
 Academic dissertation. Helsinki: University of Helsinki.
- Shalini, R., & Gupta, D. K. (2010). Utilization of pomace from apple processing industries: a
 review. *Journal of Food Science and Technology*, 47(4): 365-371.
 http://doi.org/10.1007/s13197-010-0061-x.
- 617 Symoneaux, R., Chollet, S., Patron, C., Bauduin, R., Le Quéré, J. M., & Baron, A. (2015). Prediction of Sensory Characteristics of Cider According to Their Biochemical 618 619 Composition: Use of a Central Composite Design and External Validation by Cider 620 Professionals. LWT-Food Science and Technology, 61 (1), 63–69. 621 https://doi.org/10.1016/j.lwt.2014.11.030.
- Tufariello, M., Pati, S., D'Amico, L., Bleve, G., Losito, I., & Grieco, F. (2019). Quantitative
 Issues Related to the Headspace-SPME-GC/MS Analysis of Volatile Compounds in
 Wines: the Case of Maresco Sparkling Wine. *LWT- Food Science and Technology*, 108:
 268-276. https://doi.org/10.1016/j.lwt.2019.03.063.
- Varela, C. (2016). The impact of non-*Saccharomyces* yeasts in the production of alcoholic
 beverages. *Applied Microbiology and Biotechnology*, 100(23), 9861-9874. http://doi.org/
- 628 10.1007/s00253-016-7941-6.
- Varela, P., & Ares, G. (2012). Sensory Profiling, the Blurred Line between Sensory and
 Consumer Science. A Review of Novel Methods for Product Characterization. *Food Research International*, 48 (2), 893–908. https://doi.org/10.1016/j.foodres.2012.06.037.
- 632 Volatile Compounds in Food Database, Volatile Compounds in Food Database. (2020)
 633 www.vcf-online.nl/VcfCompounds.cfm, Accessed 31th August 2020.
- Watson, B. (2013). Cider, Hard and Sweet: History, Traditions, and Making Your Own. In
 The Countryman Press.

| 636 | Wei, J., Zhang, Y., Wang, Y., Ju, H., Niu, C.,, Yue, T. (2020). Assessment of Chemical |
|-----|--|
| 637 | Composition and Sensorial Properties of Ciders Fermented with Different Non- |
| 638 | Saccharomyces Yeasts in Pure and Mixed Fermentations. International Journal of Food |
| 639 | Microbiology, 318, 108471. https://doi.org/10.1016/j.ijfoodmicro.2019.108471. |
| 640 | Ye, M., Yue, T., & Yuan, Y. (2014). Changes in the Profile of Volatile Compounds and Amino |
| 641 | Acids during Cider Fermentation Using Dessert Variety of Apples. European Food |
| 642 | Research and Technology, 239(1): 67-77. http://doi.org/10.1007/s00217-014-2204-1. |
| 643 | |
| 644 | |
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| Sample | Malic | Quinic | Ascorbic | Succinic | Total | Fructose | Glucose | Sorbitol | Sucrose | Xylose | Total | pН | Brix | Ethanol | Total CO ₂ |
|--------------|--------------|-------------|------------|------------|-------------|--------------|-------------|------------|-------------|------------|-----------------|------------|------------|------------|---------------------------|
| | acid | acid | acid | acid | acids | (g/L) | (g/L) | (g/L) | (g/L) | (g/L) | sugars | | | content | production |
| | (g/L) | (g/L) | (g/L) | (g/L) | (g/L) | | | | | | (g/L) | | | (v/v, %) | (g) |
| Juices | | | | | | | | | | | | | | | |
| AP | $8.26 \pm$ | $1.10 \pm$ | ND | ND | $9.36 \pm$ | $93.02 \pm$ | $18.56 \pm$ | $3.76 \pm$ | $47.17 \pm$ | $0.32 \pm$ | $162.83 \pm$ | $3.24 \pm$ | - | - | - |
| | 0.22 cd | 0.09 a | | | 0.31 d | 3.43 d | 0.13 a | 0.24 a | 0.72 d | 0.02 a | 4.54 b | 0.04 b | | | |
| Kr | $2.85 \pm$ | $0.93 \pm$ | ND | ND | $3.78 \pm$ | $89.86 \pm$ | $18.22 \pm$ | $5.39 \pm$ | $57.54 \pm$ | $0.33 \pm$ | $171.34 \pm$ | $4.17 \pm$ | - | - | _ |
| | 0.22 a | 0.16 a | | | 0.38 a | 15.82 cd | 0.16 a | 0.22 b | 1.21 e | 0.07 a | 17.88 bc | 0.08 d | | | |
| LM | $4.09 \pm$ | $1.22 \pm$ | ND | ND | $5.31 \pm$ | $95.30 \pm$ | $21.51 \pm$ | $9.14 \pm$ | $34.51 \pm$ | $0.22 \pm$ | $160.68 \pm$ | $3.65 \pm$ | - | - | - |
| | 0.17 b | 0.05 a | | | 0.22 b | 0.80 d | 0.23 a | 0.05 c | 1.10 c | 0.09 a | 2.49 b | 0.01 c | | | |
| LK | $6.41 \pm$ | $1.37 \pm$ | ND | ND | $7.78 \pm$ | $85.67 \pm$ | $21.82 \pm$ | $5.74 \pm$ | $39.29 \pm$ | $0.51 \pm$ | $153.03 \pm$ | $3.07 \pm$ | - | - | - |
| | 0.05 c | 0.06 a | | | 0.11 c | 6.93 c | 0.54 a | 0.07 b | 0.51 c | 0.01 b | 8.06 b | 0.01 b | | | |
| An | $18.94 \pm$ | $10.28 \pm$ | ND | ND | $29.22 \pm$ | $63.70 \pm$ | $17.57 \pm$ | $9.68 \pm$ | $20.39 \pm$ | $0.34 \pm$ | $111.68 \pm$ | $2.68 \pm$ | - | - | - |
| | 2.06 f | 2.30 b | | | 4.36 g | 2.56 a | 2.31 a | 1.53 c | 4.32 ab | 0.05 a | 10.77 a | 0.02 a | | | |
| GB | $3.25 \pm$ | $1.02 \pm$ | ND | ND | $4.27 \pm$ | $91.55 \pm$ | $18.61 \pm$ | $2.48 \pm$ | $60.04 \pm$ | $0.21 \pm$ | $172.89 \pm$ | $3.54 \pm$ | - | - | - |
| | 0.24 a | 0.04 a | | | 0.28 a | 2.11 d | 0.41 a | 0.10 a | 0.46 e | 0.01 a | 3.09 c | 0.03 c | | | |
| Lt | $8.09 \pm$ | $1.20 \pm$ | ND | ND | $9.29 \pm$ | $63.75 \pm$ | $22.96 \pm$ | $2.23 \pm$ | $33.40 \pm$ | $0.17 \pm$ | $122.51 \pm$ | $3.11 \pm$ | - | - | - |
| | 1.08 cd | 0.17 a | | | 1.25 cd | 1.07 a | 1.76 ab | 0.40 a | 3.03 c | 0.03 a | 6.29 a | 0.02 b | | | |
| Pk | $3.43 \pm$ | $0.98 \pm$ | ND | ND | $4.41 \pm$ | $108.10 \pm$ | $23.62 \pm$ | $4.29 \pm$ | $21.06 \pm$ | $0.48 \pm$ | $157.55 \pm$ | $4.09 \pm$ | - | - | - |
| | 0.11 a | 0.01 a | | | 0.12 a | 1.02 e | 0.10 b | 0.02 b | 0.09 b | 0.02 b | 1.25 b | 0.02 d | | | |
| Tr | $9.58 \pm$ | $0.89 \pm$ | ND | ND | $10.47 \pm$ | $78.77 \pm$ | $19.27 \pm$ | $4.92 \pm$ | $50.57 \pm$ | $0.13 \pm$ | $153.66 \pm$ | $3.11 \pm$ | - | - | - |
| | 0.38 d | 0.03 a | | | 0.41 e | 1.90 b | 0.97 a | 0.29 b | 1.90 d | 0.01 a | 5.07 b | 0.02 b | | | |
| Ju | $11.31 \pm$ | $0.70 \pm$ | ND | ND | $12.01 \pm$ | $91.72 \pm$ | $20.69 \pm$ | $4.69 \pm$ | $51.35 \pm$ | $0.15 \pm$ | $168.60 \pm$ | $3.11 \pm$ | - | - | - |
| | 0.34 e | 0.02 a | | | 0.36 f | 2.53 d | 1.23 a | 0.36 b | 4.53 d | 0.03 a | 8.68 bc | 0.01 b | | | |
| Hg | $7.74 \pm$ | $0.74 \pm$ | ND | ND | $8.48 \pm$ | $84.88 \pm$ | $52.85 \pm$ | $6.54 \pm$ | $15.58 \pm$ | $0.33 \pm$ | $160.18 \pm$ | $3.41 \pm$ | - | - | - |
| | 0.31 c | 0.02 a | | | 0.33 c | 1.87 c | 0.64 c | 0.87 b | 0.11 a | 0.02 a | 3.51 b | 0.02 c | | | |
| mean | 7.63 B | 1.87 B | ND | ND | 9.48 B | 85.95 B | 23.22 B | 5.23 A | 39.13 C | 0.29 A | 153.82 B | 3.38 A | | | |
| S. cereviais | e 1116 cider | S | | | | | | | | | | | | | |
| AP | $6.74 \pm$ | $0.69 \pm$ | $0.40 \pm$ | $1.29 \pm$ | 9.12 ± | ND | $0.32 \pm$ | $3.22 \pm$ | $0.14 \pm$ | $0.24 \pm$ | 3.92 ± 0.50 | 3.29 ± | 4.25 ± | $6.82 \pm$ | 5.20 ± 0.04 c |
| | 0.11 b | 0.09 a | 0.23 ab | 0.12 b | 0.43 b | | 0.09 b | 0.37 ab | 0.03 a | 0.01 b | ab | 0.01 b | 0.07 a | 0.46 ab | |
| Kr | $2.62 \pm$ | $0.51 \pm$ | $0.29 \pm$ | $1.05 \pm$ | $4.47 \pm$ | ND | $0.18 \pm$ | $3.83 \pm$ | ND | $0.50 \pm$ | 4.51 ± 0.52 | $3.74 \pm$ | $4.20 \pm$ | $7.42 \pm$ | $5.70 \pm 0.09 \text{ d}$ |
| | 0.12 a | 0.06 a | 0.07 a | 0.05 a | 0.25 a | | 0.05 a | 0.45 b | | 0.02 d | b | 0.02 d | 0.00 a | 0.16 b | |
| LM | $3.22 \pm$ | $0.69 \pm$ | $0.59 \pm$ | $1.25 \pm$ | $5.75 \pm$ | $0.11 \pm$ | 0.39 ± | $8.38 \pm$ | ND | $0.35 \pm$ | 9.23 ± 0.77 | $3.76 \pm$ | $4.45 \pm$ | $6.67 \pm$ | 4.78 ± 0.26 |
| | 0.11 a | 0.07 a | 0.33 ab | 0.05 b | 0.51 a | 0.01 a | 0.08 b | 0.67 d | | 0.01 c | d | 0.07 d | 0.07 ab | 0.42 ab | bc |

Table 1. Chemical composition of ciders fermented with two yeast strains (S. *cerevisiae* 1116 and S. *pombe* 3796) and corresponding apple juices.

| LK | $6.07 \pm$ | $0.92 \pm$ | $0.54 \pm$ | $1.37 \pm$ | $8.90 \pm$ | ND | $0.24 \pm$ | $5.80 \pm$ | ND | $0.44 \pm$ | 6.48 ± 0.79 | $3.24 \pm$ | $4.80 \pm$ | $7.25 \pm$ | $6.51 \pm 0.26 \text{ e}$ |
|------------|-------------|------------|------------|------------|-------------|------------|------------|------------|------------|------------|-----------------|------------|------------|------------|---------------------------|
| | 0.05 b | 0.13 a | 0.03 b | 0.15 bc | 0.21 b | | 0.05 a | 0.71 c | | 0.03 d | с | 0.02 b | 0.00 b | 0.50 b | |
| An | $17.73 \pm$ | $6.33 \pm$ | $0.52 \pm$ | $1.33 \pm$ | $25.91 \pm$ | ND | $0.19 \pm$ | $4.23 \pm$ | ND | $0.21 \pm$ | 4.63 ± 0.35 | $2.72 \pm$ | $6.00 \pm$ | $5.50 \pm$ | 3.77 ± 0.07 a |
| | 0.71 e | 0.40 b | 0.06 b | 0.11 bc | 1.17 d | | 0.06 a | 0.28 bc | | 0.01 b | b | 0.01 a | 0.00 c | 0.50 a | |
| GB | $3.06 \pm$ | $0.75 \pm$ | $0.32 \pm$ | $1.58 \pm$ | $5.71 \pm$ | ND | $0.23 \pm$ | $2.45 \pm$ | ND | $0.16 \pm$ | 2.84 ± 0.52 | $3.67 \pm$ | $4.05 \pm$ | $6.18 \pm$ | $4.33\pm0.11~b$ |
| | 0.12 a | 0.12 a | 0.04 a | 0.13 c | 0.28 a | | 0.12 ab | 0.36 a | | 0.04 ab | а | 0.04 d | 0.07 a | 0.24 a | |
| Lt | $7.13 \pm$ | $0.74 \pm$ | $0.45 \pm$ | $1.41 \pm$ | $9.73 \pm$ | ND | $0.35 \pm$ | $2.04 \pm$ | ND | $0.13 \pm$ | 2.52 ± 0.40 | $3.29 \pm$ | $4.30 \pm$ | $5.97 \pm$ | $4.89\pm0.06~c$ |
| | 0.26 b | 0.21 a | 0.06 a | 0.05 c | 0.56 b | | 0.11 b | 0.28 a | | 0.01 a | а | 0.03 b | 0.00 a | 0.34 a | |
| Pk | 3.16 ± | $0.62 \pm$ | $0.73 \pm$ | $1.27 \pm$ | $5.78 \pm$ | $0.23 \pm$ | $0.52 \pm$ | $3.93 \pm$ | ND | $0.82 \pm$ | 5.27 ± 0.20 | $3.85 \pm$ | $4.30 \pm$ | $6.51 \pm$ | 4.68 ± 0.05 |
| | 0.12 a | 0.04 a | 0.05 c | 0.06 b | 0.21 a | 0.05 a | 0.09 c | 0.09 b | | 0.02 e | bc | 0.02 d | 0.00 a | 0.21 ab | bc |
| Tr | $8.79 \pm$ | $0.74 \pm$ | $0.43 \pm$ | $1.63 \pm$ | $11.59 \pm$ | ND | $0.18 \pm$ | 5.11 ± | ND | $0.12 \pm$ | 5.41 ± 0.25 | $3.20 \pm$ | $4.60 \pm$ | $7.04 \pm$ | $5.04\pm0.08~c$ |
| | 0.31 c | 0.03 a | 0.11 ab | 0.08 c | 0.45 c | | 0.02 a | 0.22 c | | 0.01 a | bc | 0.03 b | 0.14 b | 0.19 b | |
| Ju | $10.49 \pm$ | $0.44 \pm$ | $0.60 \pm$ | $1.59 \pm$ | $13.12 \pm$ | ND | $0.11 \pm$ | $4.46 \pm$ | $0.19 \pm$ | $0.13 \pm$ | 4.89 ± 0.62 | $3.27 \pm$ | $4.55 \pm$ | $6.72 \pm$ | $6.06 \pm 0.09 \text{ e}$ |
| | 0.47 d | 0.07 a | 0.04 b | 0.13 c | 0.58 c | | 0.01 a | 0.59 bc | 0.01 a | 0.01 a | b | 0.01 b | 0.07 b | 0.18 ab | |
| Hg | $6.65 \pm$ | $0.43 \pm$ | $0.59 \pm$ | $1.88 \pm$ | $9.55 \pm$ | ND | $0.57 \pm$ | $6.08 \pm$ | ND | $0.27 \pm$ | 6.92 ± 1.05 | $3.52 \pm$ | $4.60 \pm$ | $7.37 \pm$ | $5.14 \pm 0.15 \text{ c}$ |
| - | 0.47 b | 0.06 a | 0.09 b | 0.08 d | 0.62 b | | 0.08 c | 0.93 c | | 0.03 b | с | 0.02 c | 0.14 b | 0.24 b | |
| mean | 6.88 B | 1.16 A | 0.51 A | 1.42 A | 9.95 B | 0.03 A | 0.31 A | 4.51 A | 0.03 A | 0.31 A | 5.18 A | 3.41 A | 4.55 A | 6.67 A | 5.10 A |
| S. pombe 3 | 796 ciders | | | | | | | | | | | | | | |
| AP | $0.95 \pm$ | $0.78 \pm$ | $0.53 \pm$ | 2.14 ± | $4.40 \pm$ | $0.75 \pm$ | $2.80 \pm$ | 3.61 ± | $0.89 \pm$ | $0.27 \pm$ | 8.32 ± 1.82 | $4.43 \pm$ | 4.15 ± | $6.08 \pm$ | 4.62 ± 0.06 b |
| | 0.06 a | 0.09 a | 0.02 b | 0.22 c | 0.17 a | 0.06 b | 0.40 b | 0.44 ab | 0.28 b | 0.01 b | bc | 0.11 bc | 0.07 a | 0.23 b | |
| Kr | $0.48 \pm$ | $0.68 \pm$ | $0.13 \pm$ | $1.81 \pm$ | 3.10 ± | $0.49 \pm$ | 1.66 ± | $4.62 \pm$ | $0.23 \pm$ | $1.08 \pm$ | 8.08 ± 0.80 | $4.62 \pm$ | $4.20 \pm$ | $6.66 \pm$ | $5.97 \pm 0.14 \text{ d}$ |
| | 0.03 a | 0.02 a | 0.01 a | 0.04 b | 0.06 a | 0.08 a | 0.32 a | 0.28 b | 0.09 a | 0.03 d | bc | 0.18 c | 0.00 a | 0.25 bc | |
| LM | $0.97 \pm$ | $0.85 \pm$ | $0.19 \pm$ | $1.55 \pm$ | $3.56 \pm$ | 0.91 ± | 2.12 ± | $9.85 \pm$ | $0.18 \pm$ | $0.35 \pm$ | $13.41 \pm$ | 4.61 ± | $4.50 \pm$ | $6.55 \pm$ | 4.31 ± 0.25 |
| | 0.15 a | 0.06 a | 0.08 a | 0.08 a | 0.59 a | 0.21 b | 0.52 ab | 0.67 d | 0.03 a | 0.01 bc | 1.44 d | 0.07 c | 0.00 b | 0.07 bc | ab |
| LK | $1.97 \pm$ | $0.91 \pm$ | $0.18 \pm$ | $1.96 \pm$ | $5.02 \pm$ | $0.38 \pm$ | $1.98 \pm$ | $6.02 \pm$ | $0.24 \pm$ | $0.41 \pm$ | 9.03 ± 0.36 | $4.39 \pm$ | $4.65 \pm$ | $6.80 \pm$ | $5.13 \pm 0.01 \text{ c}$ |
| | 0.21 bc | 0.08 a | 0.01 a | 0.24 bc | 0.30 b | 0.02 a | 0.11 ab | 0.14 c | 0.08 a | 0.01 c | с | 0.35 bc | 0.07 b | 0.44 cd | |
| An | $12.23 \pm$ | $6.16 \pm$ | $0.24 \pm$ | $1.61 \pm$ | $20.24 \pm$ | $0.27 \pm$ | $1.71 \pm$ | 4.13 ± | $0.36 \pm$ | $0.22 \pm$ | 6.69 ± 0.56 | $2.87 \pm$ | $5.75 \pm$ | $5.19 \pm$ | 4.16 ± 0.03 a |
| | 0.33 d | 0.46 b | 0.13 a | 0.05 a | 0.92 d | 0.02 a | 0.19 a | 0.27 b | 0.07 a | 0.01 a | ab | 0.03 a | 0.07 c | 0.20 a | |
| GB | $0.85 \pm$ | $0.79 \pm$ | $0.17 \pm$ | $1.86 \pm$ | $3.67 \pm$ | $0.65 \pm$ | $1.63 \pm$ | $2.44 \pm$ | $0.09 \pm$ | $0.17 \pm$ | 4.98 ± 0.82 | $4.25 \pm$ | $4.00 \pm$ | $6.25 \pm$ | 4.06 ± 0.13 a |
| | 0.16 a | 0.05 a | 0.05 a | 0.13 b | 0.32 a | 0.07 b | 0.50 a | 0.23 a | 0.01 a | 0.01 a | а | 0.10 bc | 0.00 a | 0.05 b | |
| Lt | 3.12 ± | $1.05 \pm$ | $0.40 \pm$ | $1.55 \pm$ | $6.12 \pm$ | $0.21 \pm$ | $2.99 \pm$ | $2.77 \pm$ | $0.21 \pm$ | $0.14 \pm$ | 6.32 ± 0.85 | $4.71 \pm$ | $3.95 \pm$ | $5.93 \pm$ | $4.72\pm0.23~b$ |
| | 0.52 c | 0.61 a | 0.16 ab | 0.12 a | 1.29 c | 0.08 a | 0.45 b | 0.26 a | 0.03 a | 0.03 a | ab | 0.10 c | 0.07 a | 0.45 b | |
| Pk | $1.42 \pm$ | $0.57 \pm$ | $0.72 \pm$ | $1.72 \pm$ | 4.43 ± | $1.08 \pm$ | $3.56 \pm$ | $3.72 \pm$ | $0.31 \pm$ | $0.48 \pm$ | 9.15 ± 1.11 | $5.05 \pm$ | $4.40 \pm$ | $6.52 \pm$ | $4.65 \pm 0.04 \text{ b}$ |
| | 0.09 b | 0.09 a | 0.11 b | 0.06 ab | 0.29 ab | 0.24 b | 0.63 b | 0.94 ab | 0.09 a | 0.11 c | с | 0.05 c | 0.00 ab | 0.30 bc | |
| Tr | $0.77 \pm$ | $0.72 \pm$ | $0.28 \pm$ | $1.95 \pm$ | $3.72 \pm$ | $0.68 \pm$ | $1.86 \pm$ | $4.83 \pm$ | $0.22 \pm$ | $0.12 \pm$ | 7.71 ± 1.38 | $4.59 \pm$ | $4.30 \pm$ | 6.91 ± | 5.33 ± 0.22 c |
| | 0.14 a | 0.18 a | 0.02 a | 0.12 bc | 0.34 a | 0.16 ab | 0.32 a | 0.82 bc | 0.07 a | 0.01 a | bc | 0.04 bc | 0.00 a | 0.16 c | |
| Ju | $1.89 \pm$ | $0.48 \pm$ | $0.32 \pm$ | $2.19 \pm$ | $4.88 \pm$ | $0.89 \pm$ | $1.38 \pm$ | $4.73 \pm$ | $0.09 \pm$ | $0.15 \pm$ | 7.24 ± 0.40 | $3.80 \pm$ | $4.40 \pm$ | $7.22 \pm$ | $4.87 \pm 0.15 \text{ b}$ |
| | 0.25 b | 0.02 a | 0.09 ab | 0.15 c | 0.36 ab | 0.11 b | 0.11 a | 0.16 b | 0.01 a | 0.01 a | b | 0.15 b | 0.28 ab | 0.28 d | |

| Hg | $2.42 \pm$ | $0.48 \pm$ | $0.86 \pm$ | $1.88 \pm$ | $5.64 \pm$ | $0.95 \pm$ | $3.23 \pm$ | $6.33 \pm$ | $1.02 \pm$ | $0.29 \pm$ | $11.82 \pm$ | $4.89 \pm$ | $4.50 \pm$ | $6.84 \pm$ | $5.49\pm0.05\;c$ |
|------|------------|------------|------------|------------|------------|------------|------------|------------|------------|------------|-------------|------------|------------|------------|------------------|
| | 0.51 bc | 0.05 a | 0.06 b | 0.17 bc | 0.62 b | 0.23 b | 0.39 b | 1.00 c | 0.19 b | 0.01 b | 1.82 d | 0.10 c | 0.00 b | 0.07 c | |
| mean | 2.51 A | 1.24 A | 0.35 A | 1.79 B | 5.88 A | 0.67 A | 2.56 A | 4.87 A | 0.37 B | 0.78 B | 9.24 A | 4.38 B | 4.43 A | 6.45 A | 4.85 A |

649 Results represent the mean \pm standard deviation. Apple juices in duplicate and apple ciders in four replicates. ND: not detected.

650 Statistically significant differences between cultivars within each sample type (juice, *S. cerevisiae* or *S. pombe*) are shown with lower case letters

- a-g and for the mean values are shown with upper case letters A-C (ANOVA with Tukey's test; p<0.05).
- 652 Abbreviations of cultivar numbers refer to Supplementary Table 1.

Table 2. Identification of volatiles by HS-SPME-GC-MS in unfermented apple juices and ciders fermented with *S. cerevisiae* strain 1116 and *S.*

pombe strain 3796.

| Peak number ^a | Compounds | R | 2I ^b | BP ^c | Formula | Identification ^d | Odor descriptor ^e | Abbreviat |
|--------------------------|----------------------------|--------|-----------------|-----------------|--------------------|-----------------------------|--|-----------|
| | | DB-WAX | SPB-624 | | | | | 10115 |
| Ethyl esters | | | | | | | | |
| 1 | ethyl acetate | 894 | 1203 | 43 | $C_4H_8O_2$ | MS, LRI, STD | pineapple, sweet, pungent ¹ | E_1 |
| 2 | ethyl propanoate | 957 | | 57 | $C_5H_{10}O_2$ | MS, LRI | fruity ¹ | E_2 |
| 3 | ethyl 2-methylpropanoate | 965 | 1342 | 43 | $C_6H_{12}O_2$ | MS, LRI | sweet, tropical fruit, rubber ² | E_3 |
| 4 | ethyl butanoate | 1037 | 1483 | 71 | $C_6H_{12}O_2$ | MS, LRI, STD | fruity ³ | E_4 |
| 5 | ethyl 2-methylbutanoate | 1053 | | 57 | $C_7H_{14}O_2$ | MS, LRI | fruity ¹ | E_5 |
| 6 | ethyl pentanoate | 1134 | 1643 | 88 | $C_7 H_{14} O_2$ | MS, LRI | apple, dry fish, yeast ³ | E_6 |
| 7 | ethyl hexanoate | 1235 | 1802 | 88 | $C_8H_{16}O_2$ | MS, LRI | green apple, brandy, fruity 4 | E_7 |
| 8 | ethyl octanoate | 1440 | | 88 | $C_{10}H_{20}O_2$ | MS, LRI | fruity, brandy ³ | E_8 |
| 9 | ethyl 3-hydroxybutanoate | 1527 | | 43 | $C_{6}H_{12}O_{3}$ | MS, LRI | grape, roasted nut ⁵ | E_9 |
| 10 | ethyl nonanoate | 1545 | | 88 | $C_4H_{10}O_2$ | MS, LRI | fruity ⁵ | E_10 |
| 11 | ethyl decanoate | 1647 | | 88 | $C_{12}H_{24}O_2$ | MS, LRI | brandy, burnt, fruity ^{4,6} | E_11 |
| 12 | ethyl benzoate | 1670 | | 105 | $C_9H_{10}O_2$ | MS, LRI | fruity, fat, flowery ⁵ | E_12 |
| 13 | ethyl dodecanoate | 1850 | | 88 | $C_{14}H_{28}O_2$ | MS, LRI | floral, fruity, green apple ⁷ | E_13 |
| 14 | ethyl 3-hydroxydodecanoate | 1945 | | 117 | $C_{14}H_{28}O_3$ | MS, LRI | apple brandy ⁴ | E_14 |
| Acetate esters | | | | | | | | |
| 15 | methyl acetate | 828 | | 43 | $C_3H_6O_2$ | MS, LRI | ester, green, sweet ⁵ | AE_1 |
| 16 | propyl acetate | 977 | | 43 | $C_5H_{10}O_2$ | MS, LRI | celery, floral, pear ^{4,6} | AE_2 |
| 17 | 2-methylpropyl acetate | 1014 | | 43 | $C_6H_{12}O_2$ | MS, LRI | apple, banana, floral ⁴ | AE_3 |
| 18 | butyl acetate | 1072 | 1506 | 43 | $C_6H_{12}O_2$ | MS, LRI | apple, fruit, pungent ⁶ | AE_4 |
| 19 | 3-methylbutyl acetate | 1121 | 1607 | 43 | $C_7H_{14}O_2$ | MS, LRI | apple, fruit, sweet ⁷ | AE_5 |
| 20 | pentyl acetate | 1175 | | 43 | $C_7H_{14}O_2$ | MS, LRI | banana, fruit, sweet ⁸ | AE_6 |
| 21 | hexyl acetate | 1274 | 1825 | 43 | $C_8H_{16}O_2$ | MS, LRI | fruity ⁴ | AE_7 |
| 22 | octyl acetate | 1481 | | 43 | $C_{10}H_{20}O_2$ | MS, LRI | citrus, fat, wood ⁵ | AE_8 |
| 23 | phenethyl acetate | 1831 | 2123 | 104 | $C_{10}H_{12}O_2$ | MS, LRI | floral, fruit, honey ⁸ | AE_9 |

| Other esters | | | | | | | | |
|-----------------|---------------------------------------|------|------|-----|-------------------|--------------|---|-------|
| 24 | methyl butanoate | 986 | 1362 | 43 | $C_5H_{10}O_2$ | MS, LRI, STD | fruit, ester, floral ⁷ | OE_1 |
| 25 | butyl butanoate | 1220 | 1794 | 71 | $C_8H_{16}O_2$ | MS, LRI | floral ⁴ | OE_2 |
| 26 | hexyl 2-methylpropanoate | 1347 | | 43 | $C_{10}H_{20}O_2$ | MS, LRI | fruity, apple, beer ⁶ | OE_3 |
| 27 | methyl octanoate | 1394 | | 74 | $C_9H_{18}O_2$ | MS, LRI | fruity, orange, sweet, wine ⁶ | OE_4 |
| 28 | butyl hexanoate | 1416 | | 56 | $C_{10}H_{20}O_2$ | MS, LRI | fruity, grass, green ⁷ | OE_5 |
| 29 | hexyl butanoate | 1420 | 2014 | 43 | $C_{10}H_{20}O_2$ | MS, LRI | fruity, apple, fresh ⁴ | OE_6 |
| 30 | hexyl 2-methylbutanoate | 1432 | 2054 | 57 | $C_{11}H_{22}O_2$ | MS, LRI | strawberry ⁶ | OE_7 |
| 31 | methyl decanoate | 1604 | | 74 | $C_{11}H_{22}O_2$ | MS, LRI | fresh, wine ⁶ | OE_8 |
| 32 | hexyl hexanoate | 1618 | | 43 | $C_{12}H_{24}O_2$ | MS, LRI | apple, fruity ⁸ | OE_9 |
| 33 | butyl 3-hydroxybutanoate | 1716 | | 45 | $C_8H_{16}O_3$ | MS, LRI | fruity, brandy ⁷ | OE_10 |
| 34 | diethyl benzene-1,2- dicarboxylate | 2041 | | 149 | $C_{12}H_{14}O_4$ | MS, LRI | | OE_11 |
| Aldehydes | | | | | | | | |
| 35 | acetaldehyde | 705 | 924 | 44 | C_2H_4O | MS, LRI | pungent, ripe apple ^{1,4} | Ad_1 |
| 36 | butanal | 876 | 1185 | 44 | C_4H_8O | MS, LRI | banana, pungent ⁶ | Ad_2 |
| 37 | 3-methylbutanal | 919 | 1278 | 44 | $C_5H_{10}O$ | MS, LRI | malt, pungent ⁷ | Ad_3 |
| 38 | hexanal | 1088 | 1510 | 44 | $C_6H_{12}O$ | MS, LRI, STD | grassy, green apple ⁸ | Ad_4 |
| 39 | (E)-hex-2-enal | 1221 | | 41 | $C_6H_{10}O$ | MS, LRI | green, pungent ^{1,5} | Ad_5 |
| Higher alcohols | | | | | | | | |
| 40 | 2-methylpropan-1-ol | 1095 | 1255 | 43 | $C_4H_{10}O$ | MS, LRI | apple, fusel, malt ⁶ | HA_1 |
| 41 | butan-1-ol | 1146 | 1308 | 56 | $C_4H_{10}O$ | MS, LRI | medicine, fruit ^{4,5} | HA_2 |
| 42 | 2-methylbutan-1-ol | 1207 | 1425 | 57 | $C_5H_{12}O$ | MS, LRI, STD | banana, fusel, green, malt ^{3,4,5} | HA_3 |
| 43 | 3-methylbutan-1-ol | 1211 | 1421 | 55 | $C_5H_{12}O$ | MS, LRI, STD | alcohol, nail polish ³ | HA_4 |
| 44 | pentan-1-ol | 1252 | 1470 | 42 | $C_5H_{12}O$ | MS, LRI | balsamic, fruity ⁵ | HA_5 |
| 45 | 4-methylpentan-1-ol | 1317 | | 56 | $C_6H_{14}O$ | MS, LRI | almond, toasted ³ | HA_6 |
| 46 | 2-heptanol | 1321 | | 45 | $C_7H_{16}O$ | MS, LRI | mushroom ⁴ | HA_7 |
| 47 | 3-methylpentan-1-ol | 1330 | | 56 | $C_6H_{14}O$ | MS, LRI | green, pungent, wine ^{4,6} | HA_8 |
| 48 | 1-hexanol | 1356 | 1632 | 56 | $C_6H_{14}O$ | MS, LRI, STD | green, herbaceous ⁵ | HA_9 |
| 49 | 3-octanol | 1400 | | 59 | $C_8H_{18}O$ | MS, LRI | citrus, nut, oily ^{3,5} | HA_10 |
| 50 | (<i>E</i>)-2-hexen-1-ol | 1409 | | 57 | $C_6H_{12}O$ | MS, LRI | grass ⁶ | HA_11 |
| 51 | 1-octen-3-ol | 1452 | | 57 | $C_8H_{16}O$ | MS, LRI | thyme ⁵ | HA_12 |

| 52 | (+)-(<i>3R</i> , <i>4R</i>)-3,4-epoxyhexan- 1-ol | 1456 | | 85 | $C_6H_{12}O_2$ | MS, LRI | | HA_13 |
|----------------|---|------|------|-----|--------------------|--------------|-------------------------------------|-------|
| 53 | heptan-1-ol | 1460 | | 70 | $C_7H_{16}O$ | MS, LRI | oily ^{4,5} | HA_14 |
| 54 | 6-methyl-5-hepten-2-ol | 1467 | | 95 | $C_8H_{16}O$ | MS, LRI | rose ⁵ | HA_15 |
| 55 | 2-ethylhexan-1-ol | 1495 | | 57 | $C_8H_{18}O$ | MS, LRI | citrus, green, rose ^{3,5} | HA_16 |
| 56 | 2,4,6-trimethylheptan-4-ol | 1509 | | 69 | $C_{10}H_{22}O$ | MS, LRI | - | HA_17 |
| 57 | octan-1-ol | 1563 | | 56 | $C_8H_{18}O$ | MS, LRI | chemical, metal, burnt ⁴ | HA_18 |
| 58 | butane-2.3-diol | 1582 | | 45 | $C_4H_{10}O_2$ | MS, LRI | fruity ⁷ | HA_19 |
| 59 | pentadecan-8-ol | 1559 | | 83 | $C_{15}H_{32}O$ | MS, LRI | | HA_20 |
| Ketones | - | | | | | | | |
| 60 | 4-methylpentan-2-one | 1025 | | 43 | $C_6H_{12}O$ | MS, LRI | sulfur ⁶ | K_1 |
| 61 | 3-hydroxybutan-2-one | 1291 | | 45 | $C_4H_8O_2$ | MS, LRI | buttery, fatty ⁶ | K_2 |
| 62 | 6-methylhept-5-en-2-one | 1341 | | 43 | $C_8H_{14}O$ | MS, LRI, STD | pungent ⁶ | K_3 |
| 63 | 2,6,8-trimethylnonan-4-one | 1405 | | 69 | $C_{12}H_{24}O$ | MS, LRI | | K_4 |
| Terpenes | | | | | | | | |
| 64 | α-farnesene | 1755 | | 41 | $C_{15}H_{24}$ | MS, LRI | | T_1 |
| Acetals | | | | | | | | |
| 65 | 1-ethoxy-1-methoxyethane | 845 | | 59 | $C_5H_{12}O_2$ | MS, LRI | fruity ³ | Ac_1 |
| 66 | 1,1-diethoxyethane | 897 | 1352 | 45 | $C_6H_{14}O_2$ | MS, LRI | fruity, cream ⁶ | Ac_2 |
| 67 | 1-(1-ethoxyethoxy)pentane | 1108 | 1735 | 73 | $C_{9}H_{20}O_{2}$ | MS, LRI | fruity, alcoholic ⁸ | Ac_3 |
| Hydrocarbons | | | | | | | | |
| 68 | hexane | 600 | | 57 | $C_{6}H_{14}$ | MS, LRI, STD | apple processed ⁵ | H_1 |
| Oxides | | | | | | | | |
| 69 | 2,4,5-trimethyl-1,3-dioxolane | 946 | 1359 | 43 | $C_6H_{12}O_2$ | MS, LRI | fruity, wine ⁷ | O_1 |
| Benzenes | | | | | | | | |
| 70 | 1-ethyl-3-methylbenzene | 1248 | | 105 | $C_{9}H_{12}$ | MS, LRI | | B_1 |
| 71 | benzaldehyde | 1555 | | 105 | C_7H_6O | MS, LRI | almond, cherry ⁷ | B_2 |
| Volatile acids | | | | | | | | |
| 72 | acetic acid | 1466 | 1267 | 43 | $C_2H_4O_2$ | MS, LRI | sour, vinegar-like ⁶ | Ai_1 |
| 73 | 2-methylpropanoic acid | 1588 | | 43 | $C_4H_8O_2$ | MS, LRI | | Ai_2 |
| 74 | butanoic acid | 1652 | | 60 | $C_4H_8O_2$ | MS, LRI | rancid ⁷ | Ai_3 |
| 75 | 3-methylbutanoic acid | 1694 | | 60 | $C_5H_{10}O_2$ | MS, LRI | | Ai_4 |

| 76 | pentanoic acid | 1703 | | 60 | $C_5H_{10}O_2$ | MS, LRI | | Ai_5 |
|----|----------------|------|------|----|----------------|---------|--------------------|------|
| 77 | hexanoic acid | 1873 | 1842 | 60 | $C_6H_{12}O_2$ | MS, LRI | sweat ⁸ | Ai_6 |

656 ^a Number of volatiles investigated by DB-WAX

- ^b Retention indices of volatiles investigated by DB-WAX and SPB-624
- ⁶⁵⁸ ^c BP: base peak of mass spectrum
- ⁶⁵⁹ ^d Identification, MS: mass spectrum; LRI: literature retention index; STD: standard.
- ⁶⁶⁰ ^e Odor descriptors based on literature. ¹ Li et al., 2020, ² Alberti et al., 2016, ³ Qin et al., 2018, ⁴ Luan et al., 2018, ⁵ <u>http://www.vcf-</u>
- 661 <u>online.nl/VcfCompounds.cfm</u>, ⁶ Varela, 2016, ⁷ Zió Tufariello, Pati, D'Amico, Bleve, Losito & Grieco, 2019, ⁸ Niu et al., 2019

| 663 | Table 3. Average | concentrations of volatiles | (mg/L) in c | ciders fermented | with two yeast | strains (S. | cerevisiae 1 | 116 and S. | <i>pombe</i> 3796) an | ıd |
|-----|------------------|-----------------------------|-------------|------------------|----------------|-------------|--------------|------------|-----------------------|----|
| | 0 | | | | 2 | | | | | |

664 corresponding apple juices.

| Sample | Ethyl | Acetate | Other | Total | Aldehydes | Higher | Ketones | Terpenes | Acetals | Hydrocarb | Oxides | Benzenes | Volatile |
|------------|----------------|--------------|-------------|--------------|--------------|---------------|------------|-----------------|--------------|------------|-------------|------------|-------------|
| | esters | esters | esters | esters | | alcohols | | | | on | | | acids |
| Juices | | | | | | | | | | | | | |
| AP | $263.46 \pm$ | 0.31 ± | 6.13 ± | $269.91 \pm$ | 36.11 ± | $181.23 \pm$ | $1.32 \pm$ | $10.82 \pm$ | ND | ND | $0.42 \pm$ | $1.14 \pm$ | ND |
| | 9.19 f | 0.04 a | 1.02 b | 10.25 g | 1.91 e | 1.73 e | 0.07 c | 0.10 e | | | 0.02 a | 0.02 bc | |
| Kr | $33.02 \pm$ | $22.75 \pm$ | $2.06 \pm$ | 57.83 ± | $17.81 \pm$ | $147.35 \pm$ | $0.86 \pm$ | $0.34 \pm$ | ND | ND | $0.87 \pm$ | 1.64 ± | ND |
| | 0.38 c | 0.96 b | 0.16 a | 1.50 b | 0.75 c | 7.30 d | 0.05 ab | 0.09 a | | | 0.01 b | 0.11 c | |
| LM | $93.56 \pm$ | $72.75 \pm$ | $7.49 \pm$ | $173.79 \pm$ | $32.78 \pm$ | $130.61 \pm$ | $1.27 \pm$ | $0.34 \pm$ | ND | ND | $0.40 \pm$ | $1.08 \pm$ | $0.52 \pm$ |
| | 2.10 e | 1.83 c | 0.75 b | 4.69 f | 0.99 e | 0.76 c | 0.09 c | 0.07 a | | | 0.01 a | 0.09 bc | 0.05 a |
| LK | $87.29 \pm$ | $1.61 \pm$ | 3.49 ± | $92.40 \pm$ | $38.41 \pm$ | $163.76 \pm$ | $0.59 \pm$ | ND | ND | ND | $3.01 \pm$ | $0.99 \pm$ | ND |
| | 1.29 e | 0.12 a | 0.31 a | 1.65 c | 0.61 e | 3.21 de | 0.03 a | | | | 0.06 c | 0.03 b | |
| An | $46.82 \pm$ | ND | $0.33 \pm$ | $47.15 \pm$ | $1.52 \pm$ | $88.19 \pm$ | $1.31 \pm$ | 0.98 ± 0.03 | ND | ND | ND | $0.54 \pm$ | ND |
| | 2.33 cd | | 0.00 a | 1.23 ab | 0.12 a | 2.11 a | 0.11 c | b | | | | 0.08 a | |
| GB | $9.57 \pm$ | $66.85 \pm$ | $23.72 \pm$ | $100.14 \pm$ | $6.58 \pm$ | $139.95 \pm$ | $0.84 \pm$ | $1.27 \pm$ | ND | ND | ND | $0.78 \pm$ | $0.36 \pm$ |
| | 1.21 b | 9.23 c | 2.54 c | 12.97 cd | 0.60 b | 3.86 cd | 0.08 ab | 0.02 c | | | | 0.02 ab | 0.06 a |
| Lt | $12.90 \pm$ | $65.31 \pm$ | $31.86 \pm$ | $110.07 \pm$ | $8.70 \pm$ | $116.86 \pm$ | $3.56 \pm$ | $0.76 \pm$ | ND | ND | ND | $0.99 \pm$ | ND |
| | 1.20 b | 6.59 c | 2.80 d | 10.60 d | 0.73 b | 5.99 b | 0.35 e | 0.08 b | | | | 0.09 b | |
| Pk | $13.04 \pm$ | 3.16 ± | $23.22 \pm$ | 39.43 ± | $22.38 \pm$ | $227.67 \pm$ | $0.99 \pm$ | $2.58 \pm$ | ND | ND | ND | $1.07 \pm$ | $1.08 \pm$ |
| | 0.54 b | 0.11 a | 1.07 c | 1.72 a | 0.96 d | 2.52 f | 0.05 b | 0.51 d | | | | 0.12 bc | 0.03 b |
| Tr | $2.56 \pm$ | $110.30 \pm$ | $20.87 \pm$ | $133.74 \pm$ | $6.65 \pm$ | $127.28 \pm$ | $1.14 \pm$ | ND | ND | ND | ND | $1.12 \pm$ | ND |
| | 0.08 a | 2.16 d | 1.13 c | 3.38 e | 0.11 b | 4.91 c | 0.09 bc | | | | | 0.02 bc | |
| Ju | $58.27 \pm$ | $490.40 \pm$ | $30.19 \pm$ | $578.87 \pm$ | $26.68 \pm$ | $218.47 \pm$ | $2.71 \pm$ | ND | ND | ND | $0.21 \pm$ | $1.61 \pm$ | $1.12 \pm$ |
| | 2.87 d | 18.04 e | 3.62 d | 24.53 h | 0.87 d | 15.19 f | 0.35 d | | | | 0.03 a | 0.04 c | 0.20 b |
| Hg | $53.05 \pm$ | $4.88 \pm$ | 1.69 ± | 59.62 ± | $7.22 \pm$ | 98.51 ± | $0.62 \pm$ | $1.01 \pm$ | ND | ND | ND | $1.00 \pm$ | ND |
| | 1.13 d | 0.07 a | 0.13 a | 1.33 b | 0.54 b | 2.23 a | 0.03 a | 0.08 b | | | | 0.01 b | |
| mean | 61.23 A | 76.21 A | 13.73 B | 151.18 A | 18.62 A | 149.08 A | 1.38 A | 1.65 B | ND | ND | 0.45 A | 1.09 A | 0.28 A |
| S. cerevi. | siae 1116 cide | ers | | | | | | | | | | | |
| AP | $95.08 \pm$ | $25.65 \pm$ | 12.99 ± | $133.72 \pm$ | 39.71 ± | $2543.95 \pm$ | 0.94 ± | $0.95 \pm$ | $11.48 \pm$ | $0.85 \pm$ | 8.35 ± | 2.45 ± | 41.23 ± |
| | 8.75 bc | 2.19 a | 0.42 d | 11.36 b | 5.18 b | 50.09 d | 0.09 ab | 0.10 a | 0.83 ab | 0.09 a | 0.85 a | 0.17 b | 2.86 bc |
| Kr | $107.71 \pm$ | $27.95 \pm$ | $0.84 \pm$ | $136.49 \pm$ | $60.58 \pm$ | $1279.22 \pm$ | $0.44 \pm$ | ND | $19.93 \pm$ | $1.48 \pm$ | $22.95 \pm$ | $1.69 \pm$ | $28.22 \pm$ |
| | 8.98 c | 2.72 a | 0.49 a | 12.68 b | 2.00 c | 11.90 a | 0.03 a | | 1.06 b | 0.53 ab | 4.25 c | 0.20 a | 1.44 a |
| LM | $109.60 \pm$ | $86.33 \pm$ | $1.22 \pm$ | $197.14 \pm$ | $366.41 \pm$ | $2086.84 \pm$ | $4.00 \pm$ | $0.52 \pm$ | $134.58 \pm$ | $1.05 \pm$ | $77.78 \pm$ | $2.50 \pm$ | $36.06 \pm$ |
| | 6.02 c | 4.18 cd | 0.16 a | 10.28 c | 13.92 e | 33.69 c | 0.30 c | 0.06 a | 10.18 cd | 0.07 a | 5.84 d | 0.22 b | 2.55 b |

| LK | $155.12 \pm$ | $48.94 \pm$ | $1.85 \pm$ | $206.01 \pm$ | $395.70 \pm$ | $3983.63 \pm$ | $4.12 \pm$ | ND | 129.69 ± | $1.15 \pm$ | $200.57 \pm$ | $3.20 \pm$ | $64.92 \pm$ |
|---------|---------------|--------------|-------------|--------------|--------------|---------------|-------------|------------|--------------|------------|--------------|------------|-------------|
| | 12.42 e | 4.56 b | 0.63 ab | 17.61 c | 6.93 f | 115.40 g | 0.49 c | | 11.70 c | 0.30 ab | 5.94 e | 0.12 c | 3.82 e |
| An | $76.17 \pm$ | $25.24 \pm$ | $0.94 \pm$ | $102.35 \pm$ | $42.34 \pm$ | 2683.71 ± | $0.88 \pm$ | ND | 6.15 ± | $1.27 \pm$ | $14.97 \pm$ | $1.86 \pm$ | $56.36 \pm$ |
| | 4.09 a | 1.58 a | 0.05 a | 5.72 a | 5.31 b | 50.70 de | 0.09 a | | 0.65 a | 0.34 ab | 0.79 b | 0.16 a | 2.09 d |
| GB | $137.05 \pm$ | $80.90 \pm$ | $0.68 \pm$ | $218.63 \pm$ | $309.57 \pm$ | $2133.04 \pm$ | $5.08 \pm$ | $2.01 \pm$ | $152.22 \pm$ | $1.27 \pm$ | $27.94 \pm$ | $2.48 \pm$ | $46.57 \pm$ |
| | 8.40 d | 4.05 c | 0.03 a | 12.48 c | 29.89 d | 53.65 c | 0.20 d | 0.59 b | 9.91 d | 0.35 ab | 2.95 c | 0.18 b | 2.75 с |
| Lt | $90.59 \pm$ | $82.07 \pm$ | $2.05 \pm$ | $174.71 \pm$ | $62.24 \pm$ | $2719.40 \pm$ | $1.32 \pm$ | $1.84 \pm$ | $12.29 \pm$ | $1.88 \pm$ | $4.95 \pm$ | $2.24 \pm$ | $57.57 \pm$ |
| | 7.34 b | 3.57 c | 0.13 b | 11.03 bc | 3.27 c | 75.34 e | 0.11 b | 0.38 b | 1.11 ab | 0.39 b | 0.71 a | 0.20 b | 2.58 d |
| Pk | $98.93 \pm$ | $35.09 \pm$ | $1.03 \pm$ | $135.04 \pm$ | $23.18 \pm$ | $1929.20 \pm$ | $0.42 \pm$ | $0.46 \pm$ | $7.25 \pm$ | $1.60 \pm$ | $3.57 \pm$ | $1.71 \pm$ | $29.56 \pm$ |
| | 4.97 b | 2.13 ab | 0.21 a | 7.31 b | 1.11 a | 44.35 b | 0.07 a | 0.04 a | 0.29 a | 0.29 b | 0.15 a | 0.14 a | 2.32 a |
| Tr | $102.10 \pm$ | $92.97 \pm$ | $2.37 \pm$ | $197.44 \pm$ | $25.31 \pm$ | $3037.98 \pm$ | $0.53 \pm$ | ND | $8.00 \pm$ | $1.39 \pm$ | $5.39 \pm$ | $2.45 \pm$ | $42.84 \pm$ |
| | 5.38 bc | 4.21 d | 0.40 b | 9.99 c | 0.80 a | 49.66 f | 0.04 a | | 0.15 a | 0.11 ab | 0.08 a | 0.11 b | 4.53 bc |
| Ju | $113.14 \pm$ | $181.69 \pm$ | $1.24 \pm$ | $296.07 \pm$ | $32.25 \pm$ | $2468.52 \pm$ | $0.52 \pm$ | ND | $8.44 \pm$ | $1.48 \pm$ | $9.00 \pm$ | $2.85 \pm$ | $56.17 \pm$ |
| | 2.90 c | 12.15 e | 0.21 a | 15.26 d | 2.78 ab | 31.74 d | 0.09 a | | 0.41 a | 0.21 ab | 0.89 ab | 0.19 bc | 3.27 d |
| Hg | $94.09 \pm$ | $48.93 \pm$ | $7.89 \pm$ | $150.90 \pm$ | $34.58 \pm$ | $3224.57 \pm$ | $0.34 \pm$ | ND | $8.67 \pm$ | $1.44 \pm$ | $6.59 \pm$ | $1.78 \pm$ | $40.51 \pm$ |
| • | 2.21 b | 4.91 b | 0.19 c | 7.30 b | 4.31 ab | 115.91 f | 0.05 a | | 1.76 a | 0.05 ab | 0.92 a | 0.08 a | 2.52 bc |
| mean | 107.24 C | 66.87 A | 3.00 A | 177.14 B | 126.53 C | 2553.64 C | 1.69 A | 0.53 A | 45.34 A | 1.35 A | 34.73 C | 2.29 B | 45.46 C |
| S. pomb | e 3796 ciders | | | | | | | | | | | | |
| AP | $66.23 \pm$ | $11.13 \pm$ | 1.31 ± | $78.67 \pm$ | $27.40 \pm$ | $1611.71 \pm$ | $0.38 \pm$ | $0.37 \pm$ | $8.18 \pm$ | $1.01 \pm$ | $8.67 \pm$ | $2.05 \pm$ | $16.46 \pm$ |
| | 3.66 b | 0.75 a | 0.08 ab | 4.48 a | 2.49 a | 50.38 b | 0.07 a | 0.05 a | 0.40 a | 0.15 a | 0.65 ab | 0.17 ab | 1.20 b |
| Kr | $78.11 \pm$ | $25.67 \pm$ | $0.59 \pm$ | $104.38 \pm$ | $34.95 \pm$ | $1252.22 \pm$ | $0.40 \pm$ | ND | $13.91 \pm$ | $1.79 \pm$ | $14.74 \pm$ | $1.95 \pm$ | $14.28 \pm$ |
| | 2.23 c | 1.97 bc | 0.06 a | 4.26 b | 2.63 a | 10.89 a | 0.09 a | | 2.04 a | 0.14 a | 0.97 bc | 0.12 ab | 1.35 b |
| LM | $88.29 \pm$ | $72.33 \pm$ | $1.53 \pm$ | $162.14 \pm$ | $242.49 \pm$ | $1722.54 \pm$ | 3.79 ± | ND | $85.94 \pm$ | $1.00 \pm$ | $34.22 \pm$ | $2.24 \pm$ | $20.71 \pm$ |
| | 6.15 d | 1.84 e | 0.22 ab | 8.21 d | 6.65 b | 14.01 bc | 0.36 c | | 7.07 b | 0.15 a | 1.55 e | 0.23 ab | 1.52 c |
| LK | $125.66 \pm$ | $30.86 \pm$ | $4.75 \pm$ | $161.28 \pm$ | $290.21 \pm$ | $2963.64 \pm$ | $5.87 \pm$ | ND | $110.90 \pm$ | $1.50 \pm$ | $37.45 \pm$ | $2.55 \pm$ | $59.21 \pm$ |
| | 6.53 e | 2.36 c | 0.34 d | 9.23 d | 15.39 c | 50.97 e | 0.19 d | | 11.95 c | 0.47 a | 5.03 e | 0.24 b | 2.03 f |
| An | $69.57 \pm$ | $25.61 \pm$ | $2.71 \pm$ | $97.89 \pm$ | $42.39 \pm$ | $2986.76 \pm$ | $0.91 \pm$ | ND | $8.39 \pm$ | $1.08 \pm$ | $19.92 \pm$ | $2.39 \pm$ | $33.86 \pm$ |
| | 3.97 b | 0.72 bc | 0.13 b | 4.81 b | 7.58 a | 90.09 e | 0.07 b | | 0.59 a | 0.67 a | 1.62 c | 0.21 ab | 1.97 e |
| GB | $68.24 \pm$ | $62.59 \pm$ | $10.37 \pm$ | $141.20 \pm$ | $260.27 \pm$ | $2249.04 \pm$ | $4.18 \pm$ | $1.17 \pm$ | $125.98 \pm$ | $1.28 \pm$ | $25.26 \pm$ | 2.31 ± | $34.81 \pm$ |
| | 4.20 b | 4.40 d | 0.47 e | 9.06 cd | 22.48 bc | 75.39 d | 0.32 c | 0.12 b | 11.95 c | 0.15 a | 0.76 d | 0.30 ab | 2.54 e |
| Lt | $73.04 \pm$ | $75.75 \pm$ | 3.45 ± | $152.24 \pm$ | $29.62 \pm$ | $2175.40 \pm$ | $1.13 \pm$ | $0.59 \pm$ | $10.14 \pm$ | $1.20 \pm$ | $3.66 \pm$ | $2.00 \pm$ | $26.52 \pm$ |
| | 3.93 bc | 4.55 e | 0.27 c | 8.75 d | 4.70 a | 45.09 d | 0.12 b | 0.10 a | 0.59 a | 0.49 a | 0.77 a | 0.29 ab | 1.89 d |
| Pk | $74.35 \pm$ | $21.88 \pm$ | 9.51 ± | $105.75 \pm$ | $29.54 \pm$ | $1801.25 \pm$ | $0.43 \pm$ | ND | $11.41 \pm$ | $0.84 \pm$ | $2.89 \pm$ | $2.47 \pm$ | $17.05 \pm$ |
| | 7.03 bc | 2.40 bc | 0.28 e | 9.71 bc | 6.79 a | 38.83 c | 0.05 a | | 1.11 a | 0.15 a | 0.76 a | 0.56 b | 0.79 b |
| Tr | $52.12 \pm$ | $70.60 \pm$ | $1.49 \pm$ | $124.21 \pm$ | $32.93 \pm$ | $1852.65 \pm$ | $0.38 \pm$ | ND | $20.22 \pm$ | $1.30 \pm$ | $6.64 \pm$ | $2.52 \pm$ | 11.93 ± |
| | 2.89 a | 5.56 e | 0.29 ab | 8.75 c | 6.99 a | 72.30 c | 0.06 a | | 1.49 a | 0.21 a | 0.72 a | 0.16 b | 0.90 a |
| Ju | $48.15 \pm$ | $166.36 \pm$ | $0.89 \pm$ | 215.41 ± | 30.66 ± | $1369.92 \pm$ | $12.47 \pm$ | ND | $12.64 \pm$ | $1.51 \pm$ | $14.01 \pm$ | $1.52 \pm$ | $10.90 \pm$ |
| | 2.05 a | 6.54 f | 0.19 a | 8.79 e | 3.05 a | 26.55 a | 0.33 e | | 1.40 a | 0.19 a | 1.93 bc | 0.08 a | 0.71 a |
| | | | | | | | | | | | | | |

| Hg | $96.69 \pm$ | $18.13 \pm$ | 2.44 ± | $117.26 \pm$ | $28.45 \pm$ | $2101.43 \pm$ | $0.37 \pm$ | ND | $10.27 \pm$ | $1.57 \pm$ | 2.35 ± | $1.79 \pm$ | $14.44 \pm$ |
|------|-------------|-------------|--------|--------------|-------------|---------------|------------|--------|-------------|------------|---------|------------|-------------|
| | 5.32 d | 0.89 a | 0.29 b | 6.51 c | 3.15 a | 39.11 d | 0.08 a | | 0.40 a | 0.51 a | 0.19 a | 0.11 a | 0.83 b |
| mean | 76.40 B | 52.81 A | 3.55 A | 132.77 A | 95.36 B | 2007.87 B | 2.75 B | 0.19 A | 38.00 A | 1.28 A | 15.44 B | 2.16 B | 23.65 B |

Results represent the mean \pm standard deviation. Apple juices in duplicate and apple ciders in four replicates. ND: not detected.

666 Statistically significant differences between cultivars within each sample type (juice, ciders fermented with *S. cerevisiae* 1116 or *S. pombe* 3796)

are shown with lower case letters a-h and for the mean values are shown with upper case letters A-C. (ANOVA with Tukey's test; p<0.05)

668 Abbreviations of samples refer to Supplementary Table 1.



Figure 2





