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Compositional diversity among blackcurrant (Ribes nigrum) cultivars originating from European countries

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ABSTRACT: Berries representing 21 cultivars of blackcurrant were analyzed using liquid 26 chromatographic, gas chromatographic, and mass spectrometric methods coupled with 27 multivariate models. This study pinpointed compositional variation among cultivars of 28 29 different origins cultivated in the same location during two seasons. The chemical profiles of blackcurrants varied significantly among cultivars and growing years. The key differences 30 among cultivars of Scottish, Lithuanian, and Finnish origin were in the contents of phenolic 31 32 acids (23 vs. 16 vs. 19 mg/100 g on average, respectively), mainly as 5-O-caffeoylquinic acid, 4-O-coumaroylglucose, (*E*)-coumaroyloxymethylene-glucopyranosyloxy-(*Z*)-butenenitrile, 33 34 and 1-O-feruloylglucose. The Scottish cultivars were grouped based on the 3-O-glycosides of delphinidin and cyanidin, as were the Lithuanian cultivars. Among the Finnish samples, the 35 content of myricetin 3-O-glycosides, 4-O-caffeoylglucose, 1-O-coumaroylglucose, and 4-O-36 37 coumaroylglucose were significantly different between the two green-fruited cultivars and the black-fruited cultivars. The samples from the studied years differed in the content of phenolic 38 acid derivatives, quercetin glycosides, monosaccharides and citric acid. 39 40

41 **KEYWORDS:** blackcurrant, cultivar, organic acids, phenolic compounds, sugars

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46 **INTRODUCTION**

Horticultural plants have been used for food, fiber, biofuel, medicine, and other products to 47 sustain and enhance human life in the recent years.¹⁻³ As a species of family Grossulariaceae, 48 blackcurrants (*Ribes nigrum*) are rich source of bioactive metabolites and flavor compounds, 49 including sugars, acids, and phenolic compounds.⁴⁻⁷ Some of the compounds have significant 50 physiological effects on maintenance of cardiovascular health, restriction of cancer growth, 51 control of blood glucose levels, and other physiological functions in *in vitro* models.^{8,9} This 52 leads to a commercial exploitation of blackcurrant as food products and nutritional 53 54 supplements.

55 The contents and profiles of bioactive metabolites and flavor compounds are not present constantly in R. nigrum berries. A number of previous studies have confirmed that 56 environmental factors affect the chemical composition.¹⁰⁻¹² For example, in the berries of the 57 58 cultivar 'Vertti', the concentration of phenolic compounds, especially the conjugates of hydroxycinnamic acids, was dependent on the latitude of the growing site.¹⁰ Strong 59 60 correlation with temperature and radiation was found in the content of some phenolic compounds such as delphinidin-3-O-glucoside, delphinidin-3-O-rutinoside, and myricetin-3-61 O-glucoside in 'Melalahti', 'Mortti' and 'Ola'.¹¹ The genotype is another major factor 62 influencing chemical profile of R. nigrum berries. Vagiri et al. studied blackcurrant berries of 63 Scottish, Swedish, and Russian origins, revealing large variations in polyphenols, ascorbic 64 acid, and soluble sugars among the genotypes.¹³ Mikulic-Petkovsek found that the contents of 65 acids, sugars, and main groups of phenolics varied significantly during fruit ripening among 66 blackcurrant cultivars 'Rosenthal', 'Tenah', and 'Titania'.¹⁴ Similarly, differences were 67 observed among the cultivars 'Titania', 'Triton', 'Tsema', and 'Cacanska crna'.¹⁵ The impact 68 of genotype can also be seen in juice processing where the juice produced from a single 69 cultivar maintains its typical sensory characteristics during the process.⁶ 70

71 Due to extensive industrial demand, new cultivars of blackcurrant are always requested. The main goal of commercial breeding of new cultivars usually focuses on the adaptation of 72 plants to abiotic and biotic environment, as well as their cropping potential.¹⁶⁻¹⁷ For breeders, 73 breeding is a long and exacting work, making use of previous breeding results and even the 74 achievements of the previous breeders' generations. It is thus of high importance for breeders 75 to have better knowledge about the fruit quality characteristics of cultivars. Since some 76 77 common ancestors are typically used in cultivar development, it is even possible that there are some limitations related to fruit quality in breeding populations. 78

Likewise, food industry needs fruits with specific properties to meet the requirement of 79 80 processing or to reach the target quality of final products. These may not be achieved by using cultivars traditionally grown by its raw material producers. The chemical composition 81 of blackcurrant fruits has been traditionally less emphasized when new cultivars are selected. 82 83 The previous studies on chemical profiles of blackcurrants have focused on either a limited number of compounds or only a few cultivars. Therefore, it is necessary to investigate 84 systematically the compositional difference among a collection of blackcurrant cultivars that 85 are bred and cultivated in different countries. The results of our study provide new 86 knowledge to help breeders, trade, and food industry to ensure success in providing targeted 87 88 quality of blackcurrant fruit and fruit products.

In this study, we investigated and compared the composition of twenty-one cultivars of blackcurrants originating from five different countries. All cultivars were planted in 2009 at the same location and treated with the same cultivating practice. Samples collected during two consecutive years were analyzed in order to get on an idea of the possible seasonal variation. The variations in the compositions of various phenolic compounds, simple sugars, and acids were determined using liquid chromatographic (LC), gas chromatographic (GC)

95 and mass spectrometric (MS) methods, followed by comparison of datasets with partial least 96 squares (PLS) regression models. Our aim was to pinpoint the main groups or individual 97 compounds separating different cultivars and origins. This knowledge will assist plant 98 breeding as well as providing guidance for food industry in selection of raw-materials and 99 farmers in selecting cultivars.

100

101 MATERIALS AND METHODS

Materials. Blackcurrant cultivars originating from Scotland (9 cultivars), Lithuania (5 102 cultivars), Latvia (1 cultivar), Poland (1 cultivar), and Finland (5 cultivars) were cultivated in 103 104 the test site of Natural Resources Institute (Luke) in Piikkiö, Kaarina, southwest Finland (latitude 60°23' N, longitude 22°33' E, altitude ca. 5 m). The propagation material of plants 105 was provided by the breeding institute of each cultivar, to guarantee the true-to-type of the 106 cultivars. One-year old transplants were planted in 2009 in three rows, with a distance of 4 m 107 between rows and 1 m between plants within a row. Two plants of each cultivar were planted 108 in a plot randomized in each row. Berries for the analyses were sampled from the total 109 harvest of each plot in August 2014 and 2015, each representing one replicate sample of each 110 cultivar. The soil was silt moraine rich in organic matter. Irrigation via a trickle tape, 111 fertilization and other cultivation methods were according to the Finnish standard 112 guidelines.¹⁸ The harvesting time of each cultivar was defined by experienced horticulturist, 113 the definition being based on the color, flavor, and structure of optimally ripe berries. The 114 samples collected in year 2014 were firstly stored in a freezer at -70 °C for one year, and then 115 transferred to -20 °C together with the samples harvest in 2015 for 5 months. All frozen 116 samples were then delivered from Luke to University of Turku, and stored at -20 °C for a 117 maximum period of 15 months until all analyses were completed. The information of the 118

samples is shown in Supplemental Table 1, including cultivar names, origin countries, andharvesting dates.

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Weather conditions. Data on climatic conditions were collected in the meteorological station in the Luke Kaarina test site and provided by the Finnish Meteorological Institute (Helsinki, Finland), to give information of the climatic differences between the fruit ripening periods of the two years. The main climatic factors with the one-month time interval during July 20– August 20 are shown in **Supplemental Table 4**. The time interval was chosen to cover the harvesting period of all cultivars in 2014 and all but two very late cultivars in 2015, and at least 12 days preceding the earliest harvest date.

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Dry matter content. Approximately 5 g of currant samples were weight accurately, and cut
with blade in watch-glass. The residue on the blade was rinsed into the watch-glass with
Milli-Q water. The samples were dried in the oven (Oy Santasalo-Sohlberg Ab, Helsinki,
Finland) at 105 °C overnight until their weights reached a constant value.

134

Phenolic compounds. Phenolic compounds were identified using a Waters Acquity Ultra performance liquid chromatography (UPLC) system equipped with 2996 DAD detector, an electrospray ionization interface (ESI) and a Waters Quattro Premier mass spectrometer (Waters Corp., Milford, MA). All phenolics were characterized by comparing LC retention time and typical mass fragments with reference compounds and literature.^{6,19-29} Mass spectrometry was set in both negative- and positive-ion mode, the condition of which was reported in our previous study.²⁰

142 Two methods were applied for analysis based on the types of phenolic compounds. For

anthocyanins, 5 g of frozen berries were crushed into slurry and extracted with 15 mL of 143 acidic methanol (MeOH/HCl 99:1), followed by ultra-sonication (10 min) and centrifuge 144 $(4420 \times \text{g for } 10 \text{ min})$. The extraction was carried out three times. The three supernatants were 145 combined, and the total volume was set to 50 mL with acidic methanol. The samples were 146 filtered through a 0.2 µm syringe filter before UPLC-DAD-ESI-MS analysis (Waters Corp., 147 Milford, MA, USA). The analysis of anthocyanins was conducted according to the method 148 previously reported by Mäkilä and co-workers.¹⁹ The signal of anthocyanins in the LC 149 analyses was monitored at the wavelength of 520 nm. 150

151 Other phenolic compounds were extracted from crushed materials (15 g) with 10 mL of ethyl acetate. Ultra-sonication (15 min) and centrifuge ($4420 \times g$ for 15 min) were applied in 152 the four-time extraction. The combined supernatant was evaporated at 36 °C; the residue was 153 dissolved with 3 mL of methanol, and filtered through a 0.2 µm syringe filter. Liquid 154 chromatographic separation was performed with a Phenomenex Aeris peptide XB-C18 155 column (150 \times 4.60 mm, 3.6 μ m, Torrance, CA) at room temperature. The injection volume 156 was 10 µL and the total flow was kept at 1 mL/min. The mobile phase was a combination of 157 Milli-Q water (A) and acetonitrile (B), both containing 0.1% (v/v) of formic acid. The 158 gradient applied was: 0-15 min with 8-10% solvent B, 15-20 min with 10-13% B, 20-25 159 min with 13–16% B, 25–30 min with 16–18% B, 30–35 min with 18–20% B, 35–40 min with 160 20-22% B, 40-45 min with 22-25% B, 45-50 min with 25-60% B, 50-55 min with 60-8% 161 B, 55–57 min with 8% B. The chromatograms were recorded at three different wavelengths 162 (360 nm for flavonols, 320 nm for phenolic acids, and 280 nm for flavan-3-ols and other 163 phenolic compounds). 164

165 The quantification of the phenolics was performed using a Shimadzu LC-10AT liquid 166 chromatograph system, coupled with a SPD-M20A VP photodiode array (Shimadzu Corp., 167 Kyoto, Japan). The chromatographic conditions were the same as in the corresponding

qualitative analyses. The concentration of the compounds identified was determined using an external standard method as described previously.²⁰ The compounds lacking corresponding reference standards were quantified by the calibration curves of compounds with closest structures. For instance, cyanidin 3-*O*-(6"-coumaroyl)-glucoside was quantified by the calibration curve of cyanidin 3-*O*-glucoside ($y = 3 \times 10^{-8} x + 0.0026$, $R^2 = 0.9990$). The detail information of external standards is given in Supplemental Table 6.

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Sugars and simple organic acids. 15 g of frozen berries were crushed with a T25 digital 175 176 Ultra-Turrax (IKA Werke GmbH & Co. KG, Staufen im Breisgau, Germany) and extracted with 10 mL of Milli-Q water at room temperature. The extraction was assisted with ultra-177 sonication (15 min) and centrifuge (4420× g for 15 min). After the supernatant was collected, 178 the residue was extracted with the same procedure three times. The supernatants from the 179 four times of extraction were combined and diluted with Milli-Q water to a final volume of 180 50 mL. Sugars and simple organic acids in the samples were analyzed as trimethylsilyl 181 (TMS) derivatives by Shimadzu GC-2010 equipped with flame ionization detector (Shimadzu 182 corp., Kyoto, Japan). The compounds were identified based on the retention time of reference 183 standards. A mixed internal standard, consisting of sorbitol (for sugars) and tartaric acid (for 184 acids) was used for quantification. The methods for preparation of samples and standards, as 185 well as gas chromatographic conditions, were the same as described in the previous 186 research.¹² 187

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189 Statistical Analyses. The quantitative analyses of chemical compounds were performed in 190 triplicates. The results were calculated on the base of dry weight (mg/g or 100 g of berries) 191 and expressed as mean \pm standard deviation (SD). Partial least squares (PLS) regression with 192 full cross validation was applied to determine the correlation between chemical profile and

cultivar/country of origin/growing year by using Unscrambler 10.4 (Camo Process AS, Oslo,
Norway). PLS models were established with the concentrations of compounds as the
predictors (X-data), and the cultivars (and other factors listed above) as the responses (Ydata).

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198 **RESULTS AND DISCUSSION**

Altogether, 63 chemical compounds were identified from blackcurrant berries, primarily as 199 anthocyanins (15 compounds), flavonols (19), flavan-3-ols (4), phenolic acid derivatives (14), 200 201 organic acids (4) and sugars (6). The qualitative results and chromatographs are given in **Table 1** and **Supplemental Figure 1**, respectively. In accordance with previous study,³⁰ most 202 of phenolic compounds present in blackcurrants were anthocyanins, flavonols, flavan-3-ols, 203 204 and the derivatives of hydroxycinnamic acids (caffeic acid, coumaric acid, and ferulic acid). 205 In addition to delphinidin and cyanidin derivatives as the dominant anthocyanins in the berries, the glycosides of petunidin (peak 5&6), pelargonidin (peak 8&9), peonidin (peak 206 10&11), and malvidin (peak 12&13) were detected and confirmed based on the typical MS 207 fragmentations. These minor anthocyanins were not reported in previous studies.^{22,23} 208 Presence of anthocyanins was not the only difference between black and green cultivars. 209 Some flavonols present in black cultivars were not found in the two green-fruited cultivars, 210 such as myricetin 3-O-arabinoside (peak 19), quercetin 3-O-galactoside (peak 22), quercetin 211 212 3-O-arabinoside (peak 24), isorhamnetin 3-O-(6"-malonyl)-galactoside (peak 31), myricetinhexoside-deoxyhexoside (peak 32), and myricetin aglycone (peak 29). Organic acids in 213 blackcurrants were characterized as malic acid, citric acid, quinic acid, and ascorbic acid. The 214 main sugars in blackcurrants were fructose, glucose, and sucrose. 215

Quantification of the compounds. Sum content of phenolics ranged from 598 to 2798 217 mg/100g in black cultivars and from 47 to 104 mg/100 g in green ones (Supplemental Table 218 2). It has been discussed previously that the absence of anthocyanins resulted in the lowest 219 amount of total phenolics in green cultivars.³¹ Among all black cultivars, the total content of 220 anthocyanin was 1501 ± 587 mg/100 g, which was lower than previously detected by Mattila 221 et al. $(2057 \pm 442 \text{ mg/100 g dry weight, DW})$ in 32 Finnish blackcurrant cultivars in a 222 germplasm collection of mainly traditional cultivars.³² Nour *et al.* reported that glycosides of 223 cyanidin and delphinidin (3-O-glucoside and 3-O-rutinoside) accounted for 92-97% of total 224 anthocyanins in blackcurrants.³³ Similar percentages were found in the current study. 225 Anthocyanins formed the dominating groups of the phenolics in black-fruited samples, 226 mainly as glycosylated delphinidin (34-66% of sum content of phenolics) and cyanidin (31-227 52%). The total content of flavonols was 18–60 mg/100 mg dry weight, accounting for 1–6% 228 of sum content of phenolics in black cultivars, and 37-39% in green ones. The difference 229 between black and green cultivars was also shown in the profile of flavonols. In accordance 230 with the results published by Mikkonen et al.,³⁴ myricetin glycosides was the dominant group 231 of flavonols in the black cultivars studied; however, total content of quercetin glycosides was 232 6-8 times higher than that of myricetin glycosides in the green cultivars 'Vilma' and 233 'Venny'. 234

For phenolic acids, the conjugates of coumaric acids (47–74% of total phenolic acid derivatives) were the major components in the most of the cultivars, followed by caffeic acid (17–40%) and ferulic acid (9–20%); however, the cultivars 'Ben Tron' and 'Joniniai' contained more derivatives of caffeic acids and less of coumaric acids in both years. Moreover, the monomers of flavan-3-ols were found at a total content close to 10–20 mg/100 g.

Although the contents of simple organic acids significantly differed among the cultivars, 241 citric acid accounted for 75–97% of the total content of simple acids (Supplemental table 3) 242 in accordance with previous report.³⁵ It was followed by malic acid representing 3–20% of 243 total simple acids. The highest values of ascorbic acid were found in 'Tisel' (2.0–2.5 mg/g), 244 'Joniniai' (2.2–2.5 mg/g), and 'Ben Tirran' (1.7–2.3 mg/g); however, in Finnish black 245 cultivars, ascorbic acid was found at considerably low contents ranging from 0.2 to 0.6 mg/g. 246 A small quantity of quinic acid was detected in all the samples. As the dominating sugars in 247 all blackcurrant cultivars studied, fructose and glucose contributed 48-60% and 38-47% of 248 249 total content of sugars, respectively. The concentration of fructose was higher than that of glucose in all the cultivars. Compared to fructose and glucose, sucrose was present at a lower 250 level in the black-fruited currants as suggested by Woznicki.³⁶ In this study, 'Dainiai' showed 251 significantly higher sucrose content (12 mg/g on average) than other cultivars studied. The 252 contents of simple organic acids and sugars found in the samples in the current study deviated 253 considerably from the levels reported in some blackcurrant cultivars studied in previous 254 research studies.^{13,35} This difference was likely due to the different genetic background of the 255 cultivars included in these studies. Also, the growth locations were different in these studies; 256 therefore the environmental factors may have contributed to the difference observed. 257

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259 **Comparison of blackcurrant cultivars growing in different years.** A large and significant 260 variation in chemical variables was observed within each cultivar between years 2014 and 261 2015. For phenolic compounds, two green cultivars presented significantly lower sum-262 content of phenolics than the black cultivars. A newly bred Scottish sample, 'S 18/2/23', was 263 also low in phenolics (598–745 mg/100 g of dry berries) in both years. Since annual deviation 264 was seen likely due to the response of plants to the environment, a PLS regression model was used to find the distribution of individual compounds in different years. Regarding to
phenolic compounds, 78% of the chemical variables explained 89% of the variation among
the cultivars in 7 factors in Figure 1a. Samples from 2015 showed higher total amount of
flavan-3-ols, quercetins (primarily as quercetin 3-*O*-rutinoside), kaempferols (kaempferol 3-*O*-rutinoside), isorhamnetins, and coumaric acid derivatives than berries of the year 2014.
The PLS model did not show clear correlation between years and anthocyanins or sum
content of phenolics.

For simple acids and sugars, 'Ben Tirran' had the highest content of simple acids (53 mg/g in 2014 and 52 mg/g in 2015) among all the cultivars studied. Sugars were abundant in 'Tauriai' but poor in 'Ben Finlay'. In the plot of **Figure 1b**, 64% of the chemical variables of simple acids and sugars explained 65% of the variation among the cultivars in 2 factors. Citric acid, fructose, and glucose correlated strongly with the samples collected in year 2015, which explained the higher content of total simple acids and total sugars, respectively, in this year.

In our previous research, the weather condition in the last-months of growth before 279 harvest showed special importance for blackcurrant fruit development,^{10,11} since several main 280 primary (sugars) and secondary (anthocyanins) metabolites start accumulating in the last 281 stage of ripening of blackcurrant.³⁷ In the present study, exceptionally high temperatures 282 including both maximum day time and minimum night time temperatures were observed 283 284 from mid-July to mid-August of year 2014, which was the last month before harvesting (Supplemental Table 4). Zheng et al. reported that the average temperature of July 285 correlated positively with the content of citric acid, fructose, and glucose in the Finnish 286 cultivars 'Mortti' and 'Ola', based on analysis of berry samples collected in multiple years.³⁸ 287 In our study, temperatures were higher than those in the study of Zheng et al.³⁸, and our 288 results showed the opposite: higher temperatures were related to the reduction of these sugars 289

and citric acid. The phenomenon is commonly seen in other species too. It was shown, for instance, in strawberry (*Fragaria ananassa*) fruit that sugar content was negatively correlated to the temperature during fruit development,³⁹ and high temperatures have been shown to reduce the organic acids in berries of grapevine (*Vitis vinifera*).⁴⁰ Yet, our study was not able to determine that the climatic factors resulted in the yearly deviation of chemical composition of blackcurrant berries, due to the data limited to two growing years only.

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Comparison of blackcurrant cultivars originating from different countries. PLS models 297 298 were applied to investigate the difference among the samples in order to establish correlation between individual compounds and the cultivars. The PLS plots in Figure 2a show the 299 interactions between chemical compounds and all cultivars of blackcurrants as 74% of the 300 301 chemical variables explained 65% of the variation among the cultivars in 7 factors. Sum of phenolics and total anthocyanins correlated negatively with the green cultivars ('Venny' and 302 'Vilma') along the PC1. Along with the expected color-related compounds, myricetins, 303 primarily 3-O-glucoside, 3-O-arabinoside, and the free aglycone of myricetin, also 304 represented a negative correlation with the green cultivars. Since there was only one Latvian 305 and one Polish cultivar, comparison was conducted among black-fruited cultivars of Scottish, 306 Lithuanian, and Finnish origins. 307

The Scottish cultivars generally had higher total content of phenolic acid derivatives than 308 309 the Lithuanian samples (Figure 2b; 69% of the chemical variables explained 96% of the variation among the cultivars in 6 factors). Scottish cultivars correlated strongly to the 310 derivatives of both coumaric acid (CoA) and ferulic acid (FeA), primarily as 4-O-311 coumaroylglucose (4-Co-Glu), (E&Z)-coumaroyloxymethylene-glucopyranosyloxy-(Z)-312 butenenitrile (Co-meGlu-B1&2), and 1-O-feruloylglucose (1-Fe-Glu). Positive correlations 313 of Scottish cultivars were also found with galloylcatechin (GCat) and catechin (Cat). The 314

conjugates of both caffeic acid (CaA) and coumaric acid, peonidin glycosides, flavan-3-ols 315 and ascorbic acid (AsA) were the main variables to separate the Scottish from the Finnish 316 cultivars on the first two PCs in Figure 2c (72% variation in X-data explained 96% of the 317 variation Y-data with 6 factors). Compared to the Finnish samples, the Lithuanian cultivars 318 were richer in ascorbic acid and caffeic acid derivatives, mainly as 5-O-caffeoylquinic acid 319 (5-CaQA) (Figure 2d; 66% of variation in X-data explained 97% of variation in Y-data with 320 321 5 factors). Also, higher amounts of 3-O-coumaroylquinic acid (3-CoQA), and peonidin 3-Oglucoside (Po-Glu) characterized the Lithuanian cultivars. 322

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Comparison among cultivars within Scottish origin. The nine Scottish cultivars were 324 classified into three groups as shown in the scores plot of Figure 3a based on the variation in 325 the chemical variables (87% of the chemical variables explained 70% of the variation in Y-326 data with 7 factors). Group A contained cultivars 'Ben Dorain', 'Ben Gairn', 'Ben Starav' 327 and 'Ben Finlay'. Two newly bred cultivars, 'S 18/2/23' and '9154-3', belonged to group B; 328 group C consisted of 'Ben Hope', 'Ben Tirran', and 'Ben Tron'. Since a single PLS model 329 was not able to differentiate all Scottish blackcurrants, the comparison was performed by 330 331 groups.

In Figure 3b, 76% of the chemical variables explained 98% of the variation among the 332 cultivars in 5 factors, and the cultivars in group A had higher amounts of sum-content of 333 phenolic and total anthocyanins than group B. Positive correlations of group A were found 334 with cyanidin 3-O-rutinoside (Cy-Rut), petunidin 3-O-glucoside (Pt-Glu), pelargonidin 3-O-335 glucoside (Pl-Glu), and all glycosides of delphinidin (De) identified. Group B correlated 336 mainly to 4-O-caffeoylglucose (4-Ca-Glu) and 4-Co-Glu. The blackcurrants in group C 337 contained more phenolic acid derivatives (CaA and FeA) and flavan-3-ols than those in 338 Group A (Figure 3c; 54% of the chemical variables explained 98% of the variation among 339

the cultivars with 3 factors). Many of the minor flavonols, myricetin 3-O-galactoside (My-340 Gal), quercetin 3-O-galactoside (Qu-Gal), quercetin 3-O-arabinoside (Qu-Ara) and 341 isorhamnetin 3-O-(6"-malonyl)-galactoside (Is-maGal) were not observed in the group C, 342 which also distinguished these cultivars from others (Figure 3c, Supplemental Table 5). 343 Figure 3d (84% of the variation in X-data explained 99% of the variation in Y-data with 5 344 factors) indicated that group B was low in sum content of all studied phenolics compared to 345 group C, which was mostly due to the low content of anthocyanins (including De, Cy, Pt, Pl, 346 and Po compounds) and flavonols (myricetin derivatives). 347

348 The variations within the groups A-C of Scottish cultivars observed in Figure 3 were further examined in PLS regression plots in Figure 4. 'Ben Dorain' correlated strongly to 349 citric acid (CiA), fructose (Fru), glucose (Glu), total simple organic acids, and total sugars 350 (Figure 4a; 91% of the chemical variables explained 99% of the variation among the 351 cultivars in 6 factors). 'Ben Starav' correlated positively to both sucrose (Suc) and quinic 352 acid (QuA) in the plot consisting of factor 2 and factor 4 (not present in this paper). Cyanidin 353 3-O-arabinoside (Cy-Ara) was not found only in 'Ben Gairn'; however, petunidin 3-O-354 rutinoside (Pt-Rut), epicatechin (ECat), and 4-Co-Glu were present at higher contents. 'Ben 355 Finlay' correlated only to (E)-ferulovloxymethylene-glucopyranosyloxy-(Z)-butenenitrile 356 (Fe-meGlu-B). For minor components, myricetin 3-O-rutinoside (My-Rut), and 3-O-357 coumaroylquinic acid (3-CoQA) showed negative correlations with 'Ben Gairn', but 1-O-358 359 coumaroylglucose (1-Co-Glu) correlated positively to 'Ben Gairn'.

The common difference between two cultivars in group B was that 'S18/2/23' was more abundant in citric acid, ascorbic acid, and sucrose, whereas the cultivar of '9154-3' strongly correlated to the total content of flavonols, owing to the high concentration of glycosides of quercetin (Qu), and kaempferol (Ka) (Figure 4b; 67% of the variation in X-data explained 99% of the variation in Y-data with 2 factors). Phenolic acids in 'S18/2/23' were mainly

365 present as the derivatives of caffeic acid, but more ferulic acid conjugates were found in 366 '9154-3'.

Figure 4c showed 96% of the chemical variables explained 100% of the variation among 367 the cultivars in group C with 5 factors. 'Ben Tirran' contained the highest amount of citric 368 acid and ascorbic acid. 'Ben Tron' exhibited positive correlations with most of the glycosides 369 of anthocyanidins, which explained the highest sum-content of phenolics among the samples 370 371 in group C. Yet, delphinidin 3-O-(6"-coumaroyl)-glucoside (De-coGlu) and cyanidin 3-O-(6"-coumaroyl)-glucoside (Cy-coGlu) were abundant in 'Ben Tirran'. 'Ben Tirran' was also 372 373 rich in galloylcatechin (GCat), myricetin aglycone (My agly), and ferulic acid derivatives. High concentration of total flavonols and caffeic acid derivatives correlated positively to 374 'Ben Tron', mainly due to the presence of Qu-Glu, 5-CaQA, and Ka-Gal. Moreover, 3-O-375 coumaroylquinic acids (3-CoQA), quercetin 3-O-(6"-malonyl)-glucoside (Qu-maGlu) and a 376 coumaroylquinic acid isomer (CoQA) were quantified mostly in 'Ben Hope'. 377

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Comparison of Finnish cultivars. Aside from anthocyanins, the Finnish green cultivars 379 'Venny' and 'Vilma' contained high amounts of ascorbic acid, kaempferol glycosides (Ka-380 Gal and Ka-Rut), and the derivatives of phenolic acids (4-Co-Glu, 1-Co-Glu, Co-meGlu-B2, 381 and 1-Ca-Glu) compared to black ones. Additionally, myricetin was concentrated in black 382 cultivars in form of both glycosides (3-O-glucoside, 3-O-rutinoside, deoxylhexoside, and 3-383 384 *O*-arabinoside) and aglycone (Figure 5a). The PLS model in Figure 5b presents the variation (90% in X-data) among Finnish black cultivars (99% in Y-data with 4 factors). The sum of 385 all phenolic compounds, including the main glycosides of delphinidin and myricetin, were 386 most abundant in the cultivar 'Marski'. Quercetins correlated strongly with 'Mikael' as 3-O-387 glucoside, 3-O-galactoside, and 3-O-arabinoside. 'Mortti' contained the highest levels of 388 sucrose and 3-O-coumaroylquinic acid but the lowest concentrations of cyanidins, peonidins, 389

malvidins, and total flavonols. 'Venny' and 'Vilma' shared similar compositional characteristics, which was not surprising, both being offsprings of the cultivar 'Vertti'. 'Vilma' highly correlated with the content of sucrose and (*E*)-feruloyloxymethyleneglucopyranosyloxy-(*Z*)-butenenitrile (Ca-meGlu-B), whereas 'Venny' correlated mainly with malic acid (MaA), ascorbic acid, quinic acid, quercetin 3-*O*-rutinoside, and galloylcatechin (**Figure 5c**; 92% of the variation explained 98% of the variation among the two green cultivars with 3 factors).

397

398 Comparison of Lithuanian cultivars. Lithuanian samples were grouped as displayed in Supplemental Figure 2a&b. Group A consisted of 'Almiai', 'Dainiai', and 'Gagatai', 399 presenting higher concentration of anthocyanins (mostly as De, Cy, and Po), myricetin 400 401 glycosides, and phenolic acids (FeA derivatives) than both 'Joniniai' and 'Tauriai' in group B. Among the samples in group A, 'Almiai' correlated positively to simple organic acids 402 (mainly as CiA); whereas sucrose, malic acid, and quinic acid were abundant in 'Dainiai' 403 (Supplemental Figure 2c). The highest level of total anthocyanins was present in 'Gagatai', 404 mainly owing to the high content of delphinidin 3-O-rutinoside, delphinidin 3-O-(6"-405 coumaroyl)-glucoside, and cyanidin 3-O-(6"-coumaroyl)-glucoside. This was in agreement 406 with the results reported by Rubinskiene and co-workers showing higher content of 407 anthocyanins in 'Gagatai' than in 'Joniniai' and 'Almiai'.⁴¹ In the present study, 'Almiai' 408 409 correlated negatively to the total content of both cyanidins and myricetins. 'Dainiai' contained more (E)-coumaroyloxymethylene-glucopyranosyloxy-(Z)-butenenitrile, myricetin 410 3-O-galactoside, and myricetin 3-O-(6"-malonyl)-galactoside. Supplemental Figure 2d 411 suggested that 'Joniniai' was richer in malic acid, quinic acid, and sucrose than 'Tauriai'. 412 Positive correlations were found between 'Joniniai' and both 3-O-glycoisdes and free 413 aglycones of quercetin and myricetin, as well as some minor phenolics such as epicatechin 414

and 4-*O*-caffeoylglucose. The total content of coumaric acid derivatives was higher in
'Tauriai' due to the presence of two isomers of coumaroyloxymethylene-glucopyranosyloxybutenenitrile.

418

To our best knowledge, the present study is the first one revealing systematic information 419 on compositional variation among blackcurrant cultivars originating from different countries. 420 The overall differentiation among cultivars of different origins was highlighted by the 421 concentrations of different phenolic acid derivatives, even after more than five-year of 422 423 cultivation in the same geographical location with the same climatic condition. The study also found that the contents of organic acids, sugars and phenolic acid derivatives in blackcurrants 424 correlated strongly with growing year. This may have been caused by different weather 425 conditions during fruit development. The results provide important guidelines for the 426 selection of raw materials in food and beverage processing industry. For example, cultivar 427 'Dainiai' is rich in sucrose, and high levels of ascorbic acid were found in 'Tisel', 'Joniniai', 428 and 'Ben Tirran'. 'S 18/2/23' and '9154-3' are poor sources of anthocyanins compared to 429 other black-fruited cultivars. The manufacturers can select cultivars accordingly based on the 430 requirements of their products. 431

In addition, the knowledge of variation in metabolites is essential for breeding new 432 cultivars of blackcurrants. Besides agronomic traits such as yield, fruit size and 433 434 environmental resistance, the chemical composition in fruits of new cultivars will be probably more emphasized, when more specific information of human health-related effects 435 of different compounds will be available in the future. Our results suggest that the breeding 436 programs have resulted in variation in chemical quality of currants developed in different 437 countries. The cultivars from the same country may share more similarities than those created 438 in different countries. Therefore, it would be possible for plant breeding to improve fruit 439

quality by introducing new quality characteristics from blackcurrant cultivars originatingfrom different countries.

- 442
- 443

444 ASSOCIATED CONTENT

445 Abbreviations Used

malic acid (MaA), citric acid (CiA), quinic acid (QuA), ascorbic acid (AsA), fructose 446 anomers (Fru), glucose anomers (Glu), sucrose (Sur), delphinidin 3-O-glucoside (De-Glu), 447 448 delphinidin 3-O-rutinoside (De-Rut), cyanidin 3-O-glucoside (Cy-Glu), cyanidin 3-Orutinoside (Cy-Rut), petunidin 3-O-glucoside (Pt-Glu), petunidin 3-O-rutinoside (Pt-Rut), 449 cyanidin 3-O-arabinoside (Cy-Ara), pelargonidin 3-O-glucoside (Pl-Glu), pelargonidin 3-O-450 rutinoside (Pl-Rut), peonidin 3-O-glucoside (Po-Glu), peonidin 3-O-rutinoside (Po-Rut), 451 malvidin 3-O-glucoside (Ma-Glu), malvidin 3-O-rutinoside (Ma-Rut), delphinidin 3-O-(6"-452 coumaroyl)-glucoside (De-coGlu), cyanidin 3-O-(6"-coumaroyl)-glucoside (Cy-coGlu), 453 myricetin 3-O-rutinoside (My-Rut), myricetin 3-O-galactoside (My-Gal), myricetin 3-O-454 glucoside (My-Glu), myricetin 3-O-arabinoside (My-Ara), myricetin 3-O-(6"-malonyl)-455 galactoside (My-maGal), quercetin 3-O-rutinoside (Qu-Rut), quercetin 3-O-galactoside 456 (Qu-Gal), quercetin 3-O-glucoside (Qu-Glu), quercetin 3-O-arabinoside (Qu-Ara), 457 quercetin 3-O-(6"-malonyl)-glucoside (Qu-maGlu), kaempferol 3-O-rutinoside (Ka-Rut), 458 459 kaempferol 3-O-galactoside (Ka-Gal), isorhamnetin 3-O-glucoside (Is-Glu), myricetin aglycone (My agly), kaempferol 3-O-(6"-malonyl)-glucoside (Ka-maGlu), isorhamnetin 3-460 O-(6"-malonyl)-galactoside (Is-maGal), myricetin-hexoside-deoxyhexoside (My-hex-deox), 461 isorhamnetin 3-O-(6"-malonyl)-glucoside (Is-maGlu), quercetin aglycone (Qu agly), 5-O-462 caffeoylquinic acid (5-CaQA), 4-O-caffeoylglucose (4-Ca-Glu), 1-O-caffeoylglucose (1-Ca-463 Glu), coumaroylquinic acid isomer (CoQA), 3-O-coumaroylquinic acid (3-CoQA), 4-O-464

465	coumaroylglucose (4-Co-Glu), 1-O-coumaroylglucose (1-Co-Glu), 3-O-caffeoylquinic	acid
466	(3-CaQA), feruloylglucose isomer (Fe-Glu), 1-O-feruloylglucose (1-Fe-Glu),	(E)-
467	caffeoyloxymethylene-glucopyranosyloxy-(Z)-butenenitrile (Ca-meGlu-B),	(E)-
468	coumaroyloxymethylene-glucopyranosyloxy-(Z)-butenenitrile (Co-meGlu-B1),	(Z)-
469	coumaroyloxymethylene-glucopyranosyloxy-(Z)-butenenitrile (Co-meGlu-B2),	(E)-
470	feruloyloxymethylene-glucopyranosyloxy-(Z)-butenenitrile (Fe-meGlu-B), galloylcate	echin
471	(GCat), epigalloylcatechin (EGCat), (+)-catechin (Cat), (-)-epicatechin (ECat), aureu	sidin
472	glucoside (Au-Glu).	

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478

479 Supporting Information

The supporting information is provided: (1) Chromatographs of sugars, simple organic acids 480 and phenolic compounds in cultivars of blackcurrant (Supplemental Figure 1); (2) The 481 correlation of chemical compounds with Lithuanian cultivars of blackcurrants 482 (Supplemental Figure 2); (3) Information of blackcurrant cultivars studied (Supplemental 483 Table 1); (4) Concentrations of the main groups of phenolic compounds in blackcurrants 484 (Supplemental Table 2); (5) Concentrations of simple organic acids and sugars in 485 blackcurrants (Supplemental Table 3); (6) Climatic factors recorded at the growth location 486 of blackcurrant cultivars from 20th July to 20th August of year 2014 and 2015 (Supplemental 487 Table 4); (7) Concentrations of individual compounds identified in blackcurrant cultivars 488

- 489 (Supplemental Table 5 as an Excel file attached); (8) Information of external standards used
- 490 in quantification of phenolic compounds (**Supplemental Table 6**).
- 491

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496 Author Contributions

- 497 B. Yang, O. Laaksonen, K. Linderborg, S. Karhu and Y. Tian designed the study.
- 498 Y. Tian Identification of compounds using UPLC-DAD-ESI-MS, statistical analysis, and
- 499 manuscript writing;
- 500 O. Laaksonen Identification of compounds using UPLC-DAD-ESI-MS, statistical analysis,
- and manuscript writing and revising;
- H. Haikonen Analysis of anthocyanins using HPLC-DAD under Y. Tian and O. Laaksonen
 supervising;
- A.Vanag Analysis of simple acids and sugars using GC-FID under Y. Tian and O.
 Laaksonen supervising;
- 506 H. Ejaz Analysis of flavonol glycosides using HPLC-DAD under Y. Tian and O.
 507 Laaksonen supervising;
- 508 K. Linderborg, S. Karhu, & B. Yang contributed equally to manuscript writing and revising.

509

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- 516

517 **Conflict of Interest**

- 518 The authors in this manuscript have no conflict of interest.
- 519

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645	
646	
647	Figure captions
648	Figure 1. PLS models of comparison of blackcurrant cultivars in two different growing
649	years: a. phenolic compounds (n=8662), b. sugars and simple organic acids (n=1098).
650	Legend of the scores plots: \bigcirc for the samples harvested in Year 2015, \square means the samples
651	harvested in Year 2014. In the loading plots, the growing year is in red bold italic font and the
652	identified phenolic compounds are in blue font. The full names of these compounds are
653	referred in Table 1.
654	
655	Figure 2. PLS models of comparison of blackcurrant cultivars originating from
656	different countries: based on chemical variables (n=X) a. all cultivars (n=9760), b. the black
657	cultivars (n=6560) originating from Scotland and Lithuanian, c. the black cultivars
658	originating (n=5600) from Scotland and Finland, d. the black cultivars (n=3840) originating
659	from Lithuanian and Finland. Legend of the scores plots: 🗖 for Scottish samples, 🔴 for
660	Lithuanian samples, \blacktriangle for Latvian samples, \diamondsuit for Finnish black-fruited samples, \blacktriangledown for

Finnish green-fruited samples, and * for Polish samples. In the loading plots, the origin of 661 country is in red bold italic font and the identified phenolic compounds are in blue font. The 662 full names of the compounds are referred in Table 1. 663

Figure 3. PLS models of comparison of main groups of Scottish cultivars: a. all Scottish cultivars (n=4160), b. the comparison between groups A and B (n=2720), c. the comparison between groups A and C (n=3200), d. the comparison between groups B and C (n=2400). In the loading plots, the names of cultivars and groups are in red bold italic font and the identified phenolic compounds are in blue font. The full names of the compounds are referred in **Table 1**.

671

Figure 4. Comparison of Scottish cultivars with PLS regression models based on their chemical composition: a. the comparison within group A (n=2400), b. the comparison within group B (n=960), c. the comparison within group C (n=1140). The groups are based on the model in Figure 3. The name of cultivars is in red bold italic font and the identified phenolic compounds are in blue font. The full names of the compounds are referred in Table 1.

677

Figure 5. Comparison of Finnish cultivars with PLS regression models based on their chemical composition: a. all Finnish cultivars (n=2400), b. black cultivars (n=1440), c. green cultivars (n=960). In the loading plots, the name of cultivars is in red bold italic font and the identified phenolic compounds are in blue font. The full names of compounds are referred in Table 1.

Page 29 of 42

e 29 of 42 Table 1 Identification of phenolic compounds, organic acids, and sugars in blackcurrant (*Ribes nigrum*) cultivars

No.ª	Tentative identification ^b	Abbreviation ^c	UV λ _{max} (<i>nm</i>)	[M-H] ⁻ /[M+H] ⁺ (<i>m</i> /z)	[A-H] ⁻ /[A+H] ⁺ and other ions (<i>m/z</i>)	Identification by
	Anthocyanins					
1	delphinidin 3-O-glucoside	De-Glu	276,524	463/-	301/-	MS, Standard & Literature ^{6,20-23}
2	delphinidin 3-O-rutinoside	De-Rut	276,525	609/-	301/-	MS & Literature ^{6,20-23}
3	cyanidin 3-O-glucoside	Cy-Glu	280,516	447/-	285/-	MS, Standard & Literature ^{6,20-23}
4	cyanidin 3-O-rutinoside	Cy-Rut	280,517	593/-	285/-	MS, Standard & Literature ^{6,20-23}
5	petunidin 3-O-glucoside	Pt-Glu	276,527	477/-	315/-	MS & Literature ²⁰⁻²³
6	petunidin 3-O-rutinoside	Pt-Rut	276,527	623/-	315/-	MS & Literature ²⁰⁻²³
7	cyanidin 3-O-arabinoside	Cy-Ara	280,516	417/-	285/-	MS & Literature ²⁰⁻²³
8	pelargonidin 3-O-glucoside	Pl-Glu	278,525	431/-	269/-	MS & Literature ²¹⁻²³
9	pelargonidin 3-O-rutinoside	Pl-Rut	278,525	577/-	269/-	MS & Literature ²¹⁻²³
10	peonidin 3-O-glucoside	Po-Glu	280,517	461/-	299/-	MS & Literature ²⁰⁻²³
11	peonidin 3-O-rutinoside	Po-Rut	280,517	607/-	299/-	MS & Literature ²⁰⁻²³
12	malvidin 3-O-glucoside	Ma-Glu	281,522	491/-	329/-	MS & Literature ²⁰⁻²³
13	malvidin 3-O-rutinoside	Ma-Rut	281,522	637/-	329/-	MS & Literature ²⁰⁻²³
14	delphinidin 3-O-(6"-coumaroyl)-glucoside	De-coGlu	280,530	609/-	447,301/-	MS & Literature ²⁰⁻²³
15	cyanidin 3-O-(6"-coumaroyl)-glucoside	Cy-coGlu	280,524	593/-	447,285/-	MS & Literature ²⁰⁻²³
	Flavonols					
16	myricetin 3-O-rutinoside	My-Rut	255,265(sh),355	625/627	317/481,319	MS, Standard & Literature ^{19,20,24,25}
17	myricetin 3-O-galactoside	My-Gal	255,265(sh),355	479/481	317/319	MS, Standard & Literature ^{19,20,24}
18	myricetin 3-O-glucoside	My-Glu	255,265(sh),355	479/481	317/319	MS, Standard & Literature ^{19,20,24}
19	myricetin 3-O-arabinoside	My-Ara	255,265(sh),355	449/451	317/319	MS & Literature ^{19,20,24,25}
20	myricetin 3-O-(6"-malonyl)-galactoside	My-maGal	256,266(sh),356	565/567	521,317/319	MS & Literature ^{19,20}
21	quercetin 3-O-rutinoside	Qu-Rut	255,265(sh),355	609/611	301/465,303	MS, Standard & Literature ^{19,20,24,25}
22	quercetin 3-O-galactoside	Qu-Gal	255,265(sh),355	463/465	301/303	MS, Standard & Literature ^{19,20,24}
23	quercetin 3-O-glucoside	Qu-Glu	255,265(sh),355	463/465	301/303	MS, Standard & Literature ^{19,20,24,25}
24	quercetin 3-O-arabinoside	Qu-Ara	255,266(sh),355	433/435	301/303	MS & Literature ²⁰
25	quercetin 3-O-(6"-malonyl)-glucoside	Qu-maGlu	256,266(sh),356	549/551	505,301/303	MS & Literature ^{19,20,24,25}
26	kaempferol 3-O-rutinoside	Ka-Rut	266,346	593/595	285/449,287	MS & Literature ^{19,20,24,25}
27	kaempferol 3-O-galactoside	Ka-Gal	266,346	447/449	285/287	MS & Literature ^{20,25}
28	isorhamnetin 3-O-glucoside	Is-Glu	256,265(sh),354	477/479	315/317	MS, Standard & Literature ^{19,20,24}
29	myricetin aglycone	My agly	255,266(sh),370	317/319		MS & Literature ^{19,20,24}
30	kaempferol 3-O-(6"-malonyl)-glucoside	Ka-maGlu	265,465	533/535	489,285/287	MS & Literature ^{19,20}
31	isorhamnetin 3-O-(6"-malonyl)-galactoside	Is-maGal	256,265(sh).355	563/565	519,315/317	MS & Literature ^{19,20,24}
32	myricetin-hexoside-deoxyhexoside	My-hex-deox	255,268(sh),356	625/627	317/319	MS & Literature ²⁰
33	isorhamnetin 3-O-(6"-malonyl)-glucoside	Is-maGlu	256,265(sh),355	563/565	519,315/317	MS & Literature ^{19,20}

	Journal of Agricultural and Food Chemistry					Page 30 of 42
34	quercetin aglycone	Qu agly	274,368	301/303		MS & Literature ^{19,20,24}
	Phenolic acid derivatives					
35	5-O-caffeoylquinic acid	5-CaQA	295(sh),325	353/355	191,179/377,163	MS, Standard & Literature ^{19,20,26}
36	4-O-caffeoylglucose	4-Ca-Glu	298(sh),328	341/343	179,161/365,163	MS & Literature ^{19,20,26,27}
37	1-O-caffeoylglucose	1-Ca-Glu	296(sh),324	341/343	179,161/365,163	MS & Literature ^{19,20,26,27}
38	coumaroylquinic acid isomer	CoQA	290(sh),310	337/339	191,163/361,147	MS & Literature ^{19,20,26}
39	3-O-coumaroylquinic acid	3-CoQA	292(sh),314	337/339	191,163/361,147	MS & Literature ^{19,20,26}
40	4-O-coumaroylglucose	4-Co-Glu	298(sh),314	325/327	163/349,165	MS & Literature ^{19,20,26,27}
41	1-O-coumaroylglucose	1-Co-Glu	298(sh),314	325/327	163/349,165	MS & Literature ^{19,20,26,27}
42	3-O-caffeoylquinic acid	3-CaQA	295(sh),325	353/355	191,179/377,163	MS & Literature ^{19,20,26}
43	feruloylglucose isomer	Fe-Glu	298(sh),318	355/357	193,175/379,177	MS & Literature ^{19,20,26,27}
44	1-O-feruloylglucose	1-Fe-Glu	298(sh),318	355/357	193,175/379,177	MS & Literature ^{19,20,26,27}
45	(E)-caffeoyloxymethylene-glucopyranosyloxy-(Z)-butenenitrile	Ca-meGlu-B	296(sh),329	436/438	179,135/460,276	MS & Literature ^{19,28}
46	(E)-coumaroyloxymethylene-glucopyranosyloxy-(Z)-butenenitrile	Co-meGlu-B1	290(sh),314	420/422	163,119/444,260	MS & Literature ^{19,28}
47	(Z)-coumaroyloxymethylene-glucopyranosyloxy-(Z)-butenenitrile	Co-meGlu-B2	290(sh),314	420/422	163,119/444,260	MS & Literature ^{19,28}
48	(E) - feruloyloxymethylene - glucopyranosyloxy - (Z) - but enenitrile	Fe-meGlu-B	290(sh),328	450/452	193,134/474,290	MS & Literature ^{19,27,28}
	Flavan-3-ols					
49	galloylcatechin	GCat	280	305/307		MS
50	epigalloylcatechin	EGCat	280	305/307		MS
51	(+)-catechin	Cat	280	289/291		MS, Standard & Literature ^{20,29}
52	(-)-epicatechin	ECat	280	289/291		MS, Standard & Literature ^{20,29}
	Other phenolics					
53	aureusidin glucoside	Au-Glu	280,325(sh)	447/449	285/287	MS & Literature ¹⁹
	Organic acids					
54	malic acid	MaA	-	-	-	Standard & Literature ^{12,30}
55	citric acid	CiA	-	-	-	Standard & Literature ^{12,30}
56	quinic acid	QuA	-	-	-	Standard & Literature ¹²
57	ascorbic acid	AsA	-	-	-	Standard & Literature ^{12,30}
	Sugars					
58-60	fructose anomers	Fru	-	-	-	Standard & Literature ^{12,30}
61,62	glucose anomers	Glu	-	-	-	Standard & Literature ^{12,30}
63	sucrose	Sur	-	-	-	Standard & Literature ^{12,30}

^a The number of compounds is referred in **Supplemental Figure 1**. ^b Phenolic compounds were identified using UPLC-DAD-ESI-MS with the comparison of reference standards and previous literature. Both organic acids and sugars were identified using GC-FID with internal reference standards. ^c The abbreviation of each compound is used in PLS regression models.





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Figure 1



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Figure 3





38

Page 39 of 42 Figure 4

Journal of Agricultural and Food Chemistry



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Graphic for table of contents

