### **OPEN ACCESS**

# Photofragmentation of a DNA nucleoside thymidine; valence- vs. core ionization

To cite this article: E Itälä et al 2012 J. Phys.: Conf. Ser. 388 022078

View the article online for updates and enhancements.

### **Related content**

- <u>A comparative study of dissociation of thymidine molecules following valence or core photoionization</u>
  E Itälä, M A Huels, E Rachlew et al.
- <u>Ion-induced ionization and fragmentation</u> of DNA building blocks Thomas Schlathölter, Fresia Alvarado,
- <u>Sequential multiple ionization and</u> fragmentation of SF6 induced by an intense free electron laser pulse T Osipov, L Fang, B Murphy et al.

### **Recent citations**

Sadia Bari et al.

- <u>Thymidine Decomposition Induced by</u> <u>Low-Energy Electrons and Soft X Rays</u> <u>under N2 and O2 Atmospheres</u> Elahe Alizadeh *et al* 



## IOP ebooks<sup>™</sup>

Bringing together innovative digital publishing with leading authors from the global scientific community.

Start exploring the collection-download the first chapter of every title for free.

### Photofragmentation of a DNA nucleoside thymidine; valence- vs. core ionization

### E. Itälä<sup>*a*, *b*, <sup>1</sup></sup>, K. Kooser<sup>*a*</sup>, T. Hägerth<sup>*a*</sup>, E. Rachlew<sup>*c*</sup>, M. A. Huels<sup>*d*</sup>, E. Kukk<sup>*a*</sup>

<sup>a</sup>Department of Physics and Astronomy, University of Turku, FIN-20014 Turku, Finland

<sup>b</sup>Graduate School of Materilas Research, FIN-20500 Turku, Finland

<sup>c</sup>Atomic and Molecular Physics, Royal Institute of Technology KTH, S-20691 Stockholm, Sweden

<sup>d</sup>Department of Nuclear Medicine and Radiobiology, Facility of Medicine, University of Sherbrooke, Sherbrooke, JIH

5N4 Quebec, Canada.

**Synopsis** The photofragmentation of free thymidine molecule has been studied using combined electron- and ion spectroscopy. The results show that valence- and core ionization processes produce very different fragmentation patterns and that regardless of the photon energy, the photoionization leads almost always to dissociation of the thymidine molecule.

Although a vast number of different studies concerning the fragmentation of DNA and its components have been carried out during last decades, not so much is known about the fragmentation of free nucleosides. This is partly due to the difficulties to produce intact nucleosides since they tend to crack or polymerize if one for example tries to evaporate them. Thymidine, however, has a small (~ 20K) temperature gap, where it evaporates without any significant thermal cracking. This gives an excellent opportunity to study different fragmentation processes of free thymidine molecules; photofragmentation following valence- and core ionization in this case.



**Figure 1.** Thymidine molecule ( $C_{10}H_{14}N_2O_5$ ) consists of a nucleobase, thymine (upper ring), and a sugar, deoxyribose (lower ring).

Our experiments were carried out using 50 eV and 330 eV synchrotron radiation combined with electron energy resolved photoelectron-photoion-(photoion) coinciden-ce spectroscopy. This is, to our knowledge, the first study on fragmentation of free thymidine molecules using photons instead of electrons [1, 2]. The results show several interesting features concerning the valence ionization compared to core ionization of the sample molecule. While the valence ionization causes the molecule to fragment mainly between the sugar-base bond, the core ionization is much more violent producing fragments with relatively small masses. Moreover, comparison of fragmentation between thymidine, thymine and deoxyribose shows that thymine is clearly the more stable component in thymidine molecule than deoxyribose. Especially the small fragments resulting from the core ionization of thymidine seem to originate mostly from the sugar part. This suggests that it is not necessarily the base-sugar or the sugarphosphate bonds, that are the weak links in DNA.

#### References

[1] S. Ptasińska et al. 2006 Angew. Chem. Int. Ed. 45 1893.

[2] S. Ptasińska et al. 2005 Chem. Phys. Lett. 409 270.

E-mail: ersita@utu.fi